

# trans-1,2-Dimethylcycloheptane

Inchi:	InChI=1S/C9H18/c1-8-6-4-3-5-7-9(8)2/h8-9H,3-7H2,1-2H3/t8-,9-/m0/s1
InchiKey:	CLYDEJQPKBFLJW-IUCAKERBSA-N
Formula:	C9H18
SMILES:	CC1CCCCC1C
Mol. weight [g/mol]:	126.24
CAS:	13151-50-3

## Physical Properties

Property code	Value	Unit	Source
gf	29.54	kJ/mol	Joback Method
hf	-201.27	kJ/mol	Joback Method
hfus	9.87	kJ/mol	Joback Method
hvap	35.92	kJ/mol	Joback Method
ie	10.31 ± 0.05	eV	NIST Webbook
ie	10.31	eV	NIST Webbook
log10ws	-3.00		Crippen Method
logp	3.223		Crippen Method
mcvol	126.810	ml/mol	McGowan Method
pc	2814.34	kPa	Joback Method
rinpola	941.00		NIST Webbook
tb	424.47	K	Joback Method
tc	631.20	K	Joback Method
tf	190.81	K	Joback Method
vc	0.464	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	254.42	J/mol×K	424.47	Joback Method
cpg	345.35	J/mol×K	596.74	Joback Method
cpg	328.92	J/mol×K	562.29	Joback Method
cpg	311.63	J/mol×K	527.83	Joback Method
cpg	293.45	J/mol×K	493.38	Joback Method
cpg	274.39	J/mol×K	458.92	Joback Method

cpg	360.93	J/mol×K	631.20	Joback Method
dvisc	0.0002364	Paxs	424.47	Joback Method
dvisc	0.0003181	Paxs	385.53	Joback Method
dvisc	0.0004576	Paxs	346.58	Joback Method
dvisc	0.0007219	Paxs	307.64	Joback Method
dvisc	0.0012995	Paxs	268.70	Joback Method
dvisc	0.0028552	Paxs	229.75	Joback Method
dvisc	0.0086504	Paxs	190.81	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C13151503&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C13151503&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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