

# Glutaric acid, 2,6-dimethoxyphenyl isohexyl ester

Inchi:	InChI=1S/C19H28O6/c1-14(2)8-7-13-24-17(20)11-6-12-18(21)25-19-15(22-3)9-5-10-16(
InchiKey:	OPYFHYHSDCSVON-UHFFFAOYSA-N
Formula:	C19H28O6
SMILES:	COc1cccc(OC)c1OC(=O)CCCC(=O)OCCCC(C)C
Mol. weight [g/mol]:	352.42

## Physical Properties

Property code	Value	Unit	Source
gf	-478.03	kJ/mol	Joback Method
hf	-981.22	kJ/mol	Joback Method
hfus	42.66	kJ/mol	Joback Method
hvap	84.23	kJ/mol	Joback Method
log10ws	-4.41		Crippen Method
logp	3.759		Crippen Method
mcvol	281.430	ml/mol	McGowan Method
pc	1397.50	kPa	Joback Method
rinqol	2564.00		NIST Webbook
tb	867.74	K	Joback Method
tc	1071.97	K	Joback Method
tf	529.13	K	Joback Method
vc	1.069	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	876.78	J/molxK	867.74	Joback Method
cpg	891.71	J/molxK	901.78	Joback Method
cpg	905.35	J/molxK	935.82	Joback Method
cpg	917.69	J/molxK	969.86	Joback Method
cpg	928.72	J/molxK	1003.90	Joback Method
cpg	938.43	J/molxK	1037.93	Joback Method
cpg	946.80	J/molxK	1071.97	Joback Method
dvisc	0.0003122	Paxs	529.13	Joback Method
dvisc	0.0001791	Paxs	585.57	Joback Method

dvisc	0.0001134	Paxs	642.00	Joback Method
dvisc	0.0000772	Paxs	698.43	Joback Method
dvisc	0.0000557	Paxs	754.87	Joback Method
dvisc	0.0000421	Paxs	811.31	Joback Method
dvisc	0.0000330	Paxs	867.74	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U358708&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U358708&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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