

Pentane, 3-chloro-

Other names:	1-Ethylpropyl chloride 3-Chloropentane
Inchi:	InChI=1S/C5H11Cl/c1-3-5(6)4-2/h5H,3-4H2,1-2H3
InchiKey:	CXQSCYIVCSCSES-UHFFFAOYSA-N
Formula:	C5H11Cl
SMILES:	CCC(Cl)CC
Mol. weight [g/mol]:	106.59
CAS:	616-20-6

Physical Properties

Property code	Value	Unit	Source
gf	-23.15	kJ/mol	Joback Method
hf	-167.55	kJ/mol	Joback Method
hfus	9.38	kJ/mol	Joback Method
hvap	30.72	kJ/mol	Joback Method
log10ws	-2.63		Aqueous Solubility Prediction Method
logp	2.414		Crippen Method
mcvol	93.550	ml/mol	McGowan Method
pc	3352.86	kPa	Joback Method
rinpol	706.00		NIST Webbook
rinpol	721.00		NIST Webbook
rinpol	706.00		NIST Webbook
rinpol	721.00		NIST Webbook
rinpol	695.00		NIST Webbook
ripol	878.00		NIST Webbook
ripol	892.00		NIST Webbook
ripol	900.00		NIST Webbook
ripol	880.00		NIST Webbook
tb	368.05 ± 2.00	K	NIST Webbook
tb	370.94 ± 0.30	K	NIST Webbook
tb	370.65 ± 4.00	K	NIST Webbook
tb	370.70	K	NIST Webbook
tc	527.49	K	Joback Method
tf	168.00 ± 2.00	K	NIST Webbook
tf	168.00 ± 2.00	K	NIST Webbook
vc	0.358	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	201.54	J/mol×K	527.49	Joback Method
cpg	193.96	J/mol×K	498.04	Joback Method
cpg	186.05	J/mol×K	468.59	Joback Method
cpg	177.80	J/mol×K	439.14	Joback Method
cpg	169.22	J/mol×K	409.69	Joback Method
cpg	160.28	J/mol×K	380.24	Joback Method
cpg	150.99	J/mol×K	350.79	Joback Method
dvisc	0.0078623	Pa×s	161.03	Joback Method
dvisc	0.0002823	Pa×s	350.79	Joback Method
dvisc	0.0003734	Pa×s	319.16	Joback Method
dvisc	0.0005253	Pa×s	287.54	Joback Method
dvisc	0.0008040	Pa×s	255.91	Joback Method
dvisc	0.0013876	Pa×s	224.28	Joback Method
dvisc	0.0028647	Pa×s	192.66	Joback Method
hvapt	36.50	kJ/mol	349.50	NIST Webbook

Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.40997e+01
Coeff. B	-3.01495e+03
Coeff. C	-5.27140e+01
Temperature range (K), min.	271.00
Temperature range (K), max.	395.78

Sources

NIST Webbook:

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C616206&Units=SI>

The Yaws Handbook of Vapor Pressure:

<https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure>

Crippen Method: <http://pubs.acs.org/doi/abs/10.1021/ci990307l>
Joback Method: https://en.wikipedia.org/wiki/Joback_method
Aqueous Solubility Prediction Method: <http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDataset002.xlsx>
McGowan Method: <http://link.springer.com/article/10.1007/BF02311772>

Legend

cpg:	Ideal gas heat capacity
dvisc:	Dynamic viscosity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
hvapt:	Enthalpy of vaporization at a given temperature
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
pvap:	Vapor pressure
rinpol:	Non-polar retention indices
ripol:	Polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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