

# Mestanolone

## Other names:

17«beta»-Hydroxy-17«alpha»-methyl-5«alpha»-androstan-3-one  
17-«alpha»-Methylandrostan-17-«beta»-ol-3-one  
Androstan-3-one, 17-hydroxy-17-methyl-, (5«alpha»,17«beta»)-  
Andoron  
Androstonone  
Assimil  
Ermalone  
Mesanolon  
Mestalone  
Methybol  
Methylantalon  
Methyldihydrotestosterone  
Preroide  
Tantarone  
17«alpha»-Methyl-17«beta»-hydroxy-5«alpha»-androstan-3-one  
17«alpha»-Methyl-5«alpha»-dihydrotestosterone  
17«alpha»-Methyldihydrotestosterone  
5«alpha»-Androstan-3-one, 17«beta»-hydroxy-17-methyl-  
5«alpha»-Androstane-17«alpha»-methyl-17«beta»-ol-3-one  
5«alpha»-Androstan-3-on-17«beta»-ol, 17«alpha»-methyl  
NSC 18219  
RU 143  
5A-Androstan-3-on-17B-ol, 17A-methyl

**Inchi:** InChI=1S/C20H32O2/c1-18-9-6-14(21)12-13(18)4-5-15-16(18)7-10-19(2)17(15)8-11-20(18)20

**InchiKey:** WYZDXEKUWRCKOB-PPHAMSMSSA-N

**Formula:** C20H32O2

**SMILES:** CC12CCC(=O)CC1CCC1C2CCC2(C)C1CCC2(C)O

**Mol. weight [g/mol]:** 304.47

**CAS:** 521-11-9

## Physical Properties

Property code	Value	Unit	Source
gf	1.01	kJ/mol	Joback Method
hf	-500.96	kJ/mol	Joback Method
hfus	17.51	kJ/mol	Joback Method
hvap	77.17	kJ/mol	Joback Method

log10ws	-4.98		Crippen Method
logp	4.349		Crippen Method
mcvol	256.660	ml/mol	McGowan Method
pc	1841.99	kPa	Joback Method
rinpol	2520.00		NIST Webbook
rinpol	2609.00		NIST Webbook
rinpol	2609.00		NIST Webbook
tb	852.02	K	Joback Method
tc	1089.40	K	Joback Method
tf	557.34	K	Joback Method
vc	0.961	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	929.72	J/mol×K	852.02	Joback Method
cpg	957.55	J/mol×K	891.58	Joback Method
cpg	986.09	J/mol×K	931.15	Joback Method
cpg	1015.78	J/mol×K	970.71	Joback Method
cpg	1047.06	J/mol×K	1010.27	Joback Method
cpg	1080.37	J/mol×K	1049.84	Joback Method
cpg	1116.14	J/mol×K	1089.40	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C521119&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C521119&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions

<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvac:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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