

# Tricosane, 2,6,10,14,18,22-hexamethyl

<b>Inchi:</b>	InChI=1S/C29H60/c1-24(2)14-9-16-26(5)18-11-20-28(7)22-13-23-29(8)21-12-19-27(6)17
<b>InchiKey:</b>	OKAPZOWZLTVHEP-UHFFFAOYSA-N
<b>Formula:</b>	C29H60
<b>SMILES:</b>	CC(C)CCCC(C)CCCC(C)CCCC(C)CCCC(C)CCCC(C)C
<b>Mol. weight [g/mol]:</b>	408.79

## Physical Properties

Property code	Value	Unit	Source
gf	178.66	kJ/mol	Joback Method
hf	-673.57	kJ/mol	Joback Method
hfus	49.73	kJ/mol	Joback Method
hvap	77.82	kJ/mol	Joback Method
log10ws	-10.51		Crippen Method
logp	10.694		Crippen Method
mcvol	419.470	ml/mol	McGowan Method
pc	639.63	kPa	Joback Method
rinpol	2523.00		NIST Webbook
rinpol	2523.00		NIST Webbook
rinpol	2523.00		NIST Webbook
rinpol	2523.00		NIST Webbook
rinpol	2522.00		NIST Webbook
tb	860.28	K	Joback Method
tc	1053.42	K	Joback Method
tf	326.59	K	Joback Method
vc	1.623	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1416.90	J/mol×K	860.28	Joback Method
cpg	1442.64	J/mol×K	892.47	Joback Method
cpg	1466.96	J/mol×K	924.66	Joback Method
cpg	1489.92	J/mol×K	956.85	Joback Method
cpg	1511.60	J/mol×K	989.04	Joback Method

cpg	1532.04	J/molxK	1021.23	Joback Method
cpg	1551.33	J/molxK	1053.42	Joback Method
dvisc	0.0082516	Paxs	326.59	Joback Method
dvisc	0.0009690	Paxs	415.54	Joback Method
dvisc	0.0002422	Paxs	504.49	Joback Method
dvisc	0.0000917	Paxs	593.43	Joback Method
dvisc	0.0000447	Paxs	682.38	Joback Method
dvisc	0.0000258	Paxs	771.33	Joback Method
dvisc	0.0000166	Paxs	860.28	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R213851&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R213851&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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