

# Terephthalic acid, 2,2-dichloroethyl hexyl ester

Inchi:	InChI=1S/C16H20Cl2O4/c1-2-3-4-5-10-21-15(19)12-6-8-13(9-7-12)16(20)22-11-14(17)18
InchiKey:	PLWSOSCJUKVFBC-UHFFFAOYSA-N
Formula:	C16H20Cl2O4
SMILES:	CCCCCOC(=O)c1ccc(C(=O)OCC(Cl)Cl)cc1
Mol. weight [g/mol]:	347.23

## Physical Properties

Property code	Value	Unit	Source
gf	-307.52	kJ/mol	Joback Method
hf	-674.87	kJ/mol	Joback Method
hfus	41.29	kJ/mol	Joback Method
hvap	80.84	kJ/mol	Joback Method
log10ws	-5.38		Crippen Method
logp	4.384		Crippen Method
mvol	251.900	ml/mol	McGowan Method
pc	1718.88	kPa	Joback Method
rinpol	2431.00		NIST Webbook
tb	824.14	K	Joback Method
tc	1036.24	K	Joback Method
tf	498.18	K	Joback Method
vc	0.964	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	700.68	J/molxK	824.14	Joback Method
cpg	713.70	J/molxK	859.49	Joback Method
cpg	725.67	J/molxK	894.84	Joback Method
cpg	736.63	J/molxK	930.19	Joback Method
cpg	746.58	J/molxK	965.54	Joback Method
cpg	755.55	J/molxK	1000.89	Joback Method
cpg	763.55	J/molxK	1036.24	Joback Method
dvisc	0.0006699	Paxs	498.18	Joback Method
dvisc	0.0003768	Paxs	552.51	Joback Method

dvisc	0.0002349	Paxs	606.83	Joback Method
dvisc	0.0001583	Paxs	661.16	Joback Method
dvisc	0.0001132	Paxs	715.49	Joback Method
dvisc	0.0000849	Paxs	769.81	Joback Method
dvisc	0.0000662	Paxs	824.14	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U356246&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U356246&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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