

Butane, 2,3-dichloro-

Other names:	2,3-Dichlorobutane D,l-2,3-dichlorobutane
Inchi:	InChI=1S/C4H8Cl2/c1-3(5)4(2)6/h3-4H,1-2H3
InchiKey:	RMISVOPUIFJTEO-UHFFFAOYSA-N
Formula:	C4H8Cl2
SMILES:	CC(Cl)C(C)Cl
Mol. weight [g/mol]:	127.01
CAS:	7581-97-7

Physical Properties

Property code	Value	Unit	Source
gf	-45.94	kJ/mol	Joback Method
hf	-167.93	kJ/mol	Joback Method
hfus	7.46	kJ/mol	Joback Method
hvap	32.49	kJ/mol	Joback Method
log10ws	-2.70		Aqueous Solubility Prediction Method
logp	2.241		Crippen Method
mcvol	91.700	ml/mol	McGowan Method
pc	3628.97	kPa	Joback Method
rinpol	761.00		NIST Webbook
rinpol	761.00		NIST Webbook
rinpol	763.00		NIST Webbook
rinpol	761.00		NIST Webbook
tb	364.90	K	Joback Method
tc	555.91	K	Joback Method
tf	164.68	K	Joback Method
vc	0.345	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	183.87	J/mol×K	555.91	Joback Method
cpg	177.64	J/mol×K	524.07	Joback Method

cpg	171.09	J/mol×K	492.24	Joback Method
cpg	164.21	J/mol×K	460.40	Joback Method
cpg	157.00	J/mol×K	428.57	Joback Method
cpg	149.43	J/mol×K	396.73	Joback Method
cpg	141.51	J/mol×K	364.90	Joback Method
dvisc	0.0130687	Pa×s	164.68	Joback Method
dvisc	0.0003287	Pa×s	364.90	Joback Method
dvisc	0.0004458	Pa×s	331.53	Joback Method
dvisc	0.0006475	Pa×s	298.16	Joback Method
dvisc	0.0010331	Pa×s	264.79	Joback Method
dvisc	0.0018860	Pa×s	231.42	Joback Method
dvisc	0.0042176	Pa×s	198.05	Joback Method
hvapt	39.60	kJ/mol	318.00	NIST Webbook

Sources

Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
KDB:	https://www.cheric.org/research/kdb/hcprop/showprop.php?cmpid=1605
Aqueous Solubility Prediction Method:	http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDataset002.xlsx
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C7581977&Units=SI

Legend

cpg:	Ideal gas heat capacity
dvisc:	Dynamic viscosity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
hvapt:	Enthalpy of vaporization at a given temperature
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature

tc: Critical Temperature
tf: Normal melting (fusion) point
vc: Critical Volume

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