

# Cyclohexene, 3-methyl-6-(1-methylethyl)-

<b>Other names:</b>	p-Menth-2-ene 2-Menthene 3-Isopropyl-6-methylcyclohexene 3-Isopropyl-6-methyl-1-cyclohexene «DELTA»
<b>Inchi:</b>	InChI=1S/C10H18/c1-8(2)10-6-4-9(3)5-7-10/h4,6,8-10H,5,7H2,1-3H3
<b>InchiKey:</b>	WHNGPXQYYRWQAS-UHFFFAOYSA-N
<b>Formula:</b>	C10H18
<b>SMILES:</b>	CC1C=CC(C(C)C)CC1
<b>Mol. weight [g/mol]:</b>	138.25
<b>CAS:</b>	5256-65-5

## Physical Properties

Property code	Value	Unit	Source
gf	77.58	kJ/mol	Joback Method
hf	-163.25	kJ/mol	Joback Method
hfus	12.26	kJ/mol	Joback Method
hvap	37.88	kJ/mol	Joback Method
log10ws	-3.03		Crippen Method
logp	3.245		Crippen Method
mcvol	136.600	ml/mol	McGowan Method
pc	2597.78	kPa	Joback Method
rinpol	1004.00		NIST Webbook
rinpol	1004.00		NIST Webbook
rinpol	982.00		NIST Webbook
rinpol	1009.00		NIST Webbook
rinpol	985.00		NIST Webbook
rinpol	982.00		NIST Webbook
rinpol	985.00		NIST Webbook
ripol	1091.00		NIST Webbook
tb	441.80	K	Joback Method
tc	644.93	K	Joback Method
tf	191.36	K	Joback Method
vc	0.507	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	285.08	J/molxK	441.80	Joback Method
cpg	304.25	J/molxK	475.65	Joback Method
cpg	322.48	J/molxK	509.51	Joback Method
cpg	339.82	J/molxK	543.36	Joback Method
cpg	356.27	J/molxK	577.22	Joback Method
cpg	371.86	J/molxK	611.07	Joback Method
cpg	386.61	J/molxK	644.93	Joback Method
dvisc	0.0058238	Paxs	191.36	Joback Method
dvisc	0.0021384	Paxs	233.10	Joback Method
dvisc	0.0010645	Paxs	274.84	Joback Method
dvisc	0.0006369	Paxs	316.58	Joback Method
dvisc	0.0004295	Paxs	358.32	Joback Method
dvisc	0.0003145	Paxs	400.06	Joback Method
dvisc	0.0002442	Paxs	441.80	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C5256655&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C5256655&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient

<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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