

«delta»-Fenchene

Inchi:	InChI=1S/C10H16/c1-9(2)7-10(3)5-4-8(9)6-10/h4-5,8H,6-7H2,1-3H3
InchiKey:	FUIDRYCKEXJNOK-UHFFFAOYSA-N
Formula:	C10H16
SMILES:	CC12C=CC(C1)C(C)(C)C2
Mol. weight [g/mol]:	136.23

Physical Properties

Property code	Value	Unit	Source
gf	153.99	kJ/mol	Joback Method
hf	-42.37	kJ/mol	Joback Method
hfus	5.52	kJ/mol	Joback Method
hvap	35.53	kJ/mol	Joback Method
log10ws	-2.93		Crippen Method
logp	2.999		Crippen Method
mcvol	125.740	ml/mol	McGowan Method
pc	3086.42	kPa	Joback Method
rinpol	869.00		NIST Webbook
rinpol	869.00		NIST Webbook
rinpol	864.00		NIST Webbook
tb	440.92	K	Joback Method
tc	658.12	K	Joback Method
tf	279.14	K	Joback Method
vc	0.482	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	273.54	J/molxK	440.92	Joback Method
cpg	292.70	J/molxK	477.12	Joback Method
cpg	310.04	J/molxK	513.32	Joback Method
cpg	325.79	J/molxK	549.52	Joback Method
cpg	340.20	J/molxK	585.72	Joback Method
cpg	353.50	J/molxK	621.92	Joback Method
cpg	365.95	J/molxK	658.12	Joback Method

Sources

McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=R299984&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.cheméo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvp:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rlnol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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