

# Benzene, 1-fluoro-4-methyl-

<b>Other names:</b>	1-FLUORO-4-METHYLBENZENE 1-Methyl-4-fluorobenzene 4-FLUOROTOLUENE NSC 8861 Toluene, 4-fluoro Toluene, p-fluoro- p-Fluoromethylbenzene p-Fluorotoluene para-Fluorotoluene
<b>Inchi:</b>	InChI=1S/C7H7F/c1-6-2-4-7(8)5-3-6/h2-5H,1H3
<b>InchiKey:</b>	WRWPPGUCZBJXKX-UHFFFAOYSA-N
<b>Formula:</b>	C7H7F
<b>SMILES:</b>	Cc1ccc(F)cc1
<b>Mol. weight [g/mol]:</b>	110.13
<b>CAS:</b>	352-32-9

## Physical Properties

Property code	Value	Unit	Source
affp	763.80	kJ/mol	NIST Webbook
basg	736.10	kJ/mol	NIST Webbook
chl	-3777.90	kJ/mol	NIST Webbook
chl	-3747.20 ± 0.71	kJ/mol	NIST Webbook
gf	-83.97	kJ/mol	Joback Method
hf	-147.50 ± 0.08	kJ/mol	NIST Webbook
hfl	-186.90 ± 0.08	kJ/mol	NIST Webbook
hfus	10.62	kJ/mol	Joback Method
hvap	39.50	kJ/mol	NIST Webbook
hvap	39.40 ± 0.08	kJ/mol	NIST Webbook
hvap	39.47	kJ/mol	NIST Webbook
ie	8.79 ± 0.01	eV	NIST Webbook
ie	8.79 ± 0.01	eV	NIST Webbook
ie	8.79 ± 0.01	eV	NIST Webbook
log10ws	-2.28		Crippen Method
logp	2.134		Crippen Method
mcvol	87.500	ml/mol	McGowan Method
pc	3815.10	kPa	Joback Method
rinpol	787.00		NIST Webbook

rmpol	770.40		NIST Webbook
rmpol	756.00		NIST Webbook
rmpol	775.00		NIST Webbook
rmpol	777.70		NIST Webbook
rmpol	777.70		NIST Webbook
rmpol	775.00		NIST Webbook
rmpol	775.00		NIST Webbook
sl	237.11	J/molxK	NIST Webbook
tb	389.00	K	NIST Webbook
tb	389.80	K	NIST Webbook
tb	389.80	K	NIST Webbook
tc	589.50	K	NIST Webbook
tf	215.60 ± 0.20	K	NIST Webbook
tt	216.48 ± 0.01	K	NIST Webbook
vc	0.338	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	201.85	J/molxK	591.59	Joback Method
cpg	177.14	J/molxK	491.04	Joback Method
cpg	185.85	J/molxK	524.56	Joback Method
cpg	194.09	J/molxK	558.07	Joback Method
cpg	147.93	J/molxK	390.49	Joback Method
cpg	158.20	J/molxK	424.01	Joback Method
cpg	167.93	J/molxK	457.52	Joback Method
cpl	171.17	J/molxK	298.15	NIST Webbook
cpl	172.30	J/molxK	298.15	NIST Webbook
cpl	173.70	J/molxK	298.15	NIST Webbook
hfust	7.65	kJ/mol	213.00	NIST Webbook
hfust	9.35	kJ/mol	216.49	NIST Webbook
hfust	9.35	kJ/mol	216.50	NIST Webbook
hfust	9.35	kJ/mol	216.50	NIST Webbook
hfust	8.80	kJ/mol	215.55	NIST Webbook
hvapt	37.00	kJ/mol	384.50	NIST Webbook
hvapt	34.08	kJ/mol	389.80	NIST Webbook
sfust	43.19	J/molxK	216.49	NIST Webbook
sfust	41.00	J/molxK	215.55	NIST Webbook

# Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.44160e+01
Coeff. B	-3.32702e+03
Coeff. C	-5.02270e+01
Temperature range (K), min.	285.71
Temperature range (K), max.	415.65

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/T + C \cdot \ln(T) + D \cdot T^2$
Coeff. A	7.64576e+01
Coeff. B	-7.14432e+03
Coeff. C	-9.12454e+00
Coeff. D	6.06129e-06
Temperature range (K), min.	340.15
Temperature range (K), max.	590.48

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>KDB:</b>	<a href="https://www.thermo.com/files/research/kdb/mol/mol1707.mol">https://www.thermo.com/files/research/kdb/mol/mol1707.mol</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C352329&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C352329&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>KDB Vapor Pressure Data:</b>	<a href="https://www.thermo.com/research/kdb/hcprop/showprop.php?cmpid=1707">https://www.thermo.com/research/kdb/hcprop/showprop.php?cmpid=1707</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemo.com/doc/models/crippen_log10ws">https://www.chemo.com/doc/models/crippen_log10ws</a>

## Legend

**affp:** Proton affinity

<b>basg:</b>	Gas basicity
<b>chl:</b>	Standard liquid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>cpl:</b>	Liquid phase heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfl:</b>	Liquid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hfust:</b>	Enthalpy of fusion at a given temperature
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpol:</b>	Non-polar retention indices
<b>sfust:</b>	Entropy of fusion at a given temperature
<b>sl:</b>	Liquid phase molar entropy at standard conditions
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>tt:</b>	Triple Point Temperature
<b>vc:</b>	Critical Volume

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