

# 3,4-Dimethoxybenzylamine

<b>Other names:</b>	Veratrylamine Benzenemethanamine, 3,4-dimethoxy- Benzylamine, 3,4-(dimethoxy)-
<b>Inchi:</b>	InChI=1S/C9H13NO2/c1-11-8-4-3-7(6-10)5-9(8)12-2/h3-5H,6,10H2,1-2H3
<b>InchiKey:</b>	DIVNUTGTTIRPQA-UHFFFAOYSA-N
<b>Formula:</b>	C9H13NO2
<b>SMILES:</b>	COc1ccc(CN)cc1OC
<b>Mol. weight [g/mol]:</b>	167.21
<b>CAS:</b>	5763-61-1

## Physical Properties

Property code	Value	Unit	Source
gf	-25.50	kJ/mol	Joback Method
hf	-246.15	kJ/mol	Joback Method
hfus	19.90	kJ/mol	Joback Method
hvap	54.69	kJ/mol	Joback Method
log10ws	-2.02		Crippen Method
logp	1.163		Crippen Method
mcvol	135.630	ml/mol	McGowan Method
pc	3235.66	kPa	Joback Method
tb	559.33	K	Joback Method
tc	776.09	K	Joback Method
tf	370.37	K	Joback Method
vc	0.496	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	318.11	J/molxK	559.33	Joback Method
cpg	330.99	J/molxK	595.46	Joback Method
cpg	343.27	J/molxK	631.58	Joback Method
cpg	354.93	J/molxK	667.71	Joback Method
cpg	365.96	J/molxK	703.83	Joback Method
cpg	376.37	J/molxK	739.96	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	429.20	K	1.60	NIST Webbook
tbrp	393.20	K	0.40	NIST Webbook

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C5763611&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C5763611&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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