

# Benzene, 1-chloro-4-(1-methylethyl)-

<b>Other names:</b>	1-Chloro-4-isopropylbenzene 2-(p-Chlorophenyl)propane 4-Chloroisopropylbenzene 4-chlorocumene Cumene, p-chloro- p-Chlorocumene
<b>Inchi:</b>	InChI=1S/C9H11Cl/c1-7(2)8-3-5-9(10)6-4-8/h3-7H,1-2H3
<b>InchiKey:</b>	FHBSIIZALGOVLM-UHFFFAOYSA-N
<b>Formula:</b>	C9H11Cl
<b>SMILES:</b>	CC(C)c1ccc(Cl)cc1
<b>Mol. weight [g/mol]:</b>	154.64
<b>CAS:</b>	2621-46-7

## Physical Properties

Property code	Value	Unit	Source
gf	113.31	kJ/mol	Joback Method
hf	-25.05	kJ/mol	Joback Method
hfl	-77.90 ± 1.80	kJ/mol	NIST Webbook
hfus	13.39	kJ/mol	Joback Method
hvap	42.56	kJ/mol	Joback Method
log10ws	-3.34		Crippen Method
logp	3.463		Crippen Method
mcvol	126.150	ml/mol	McGowan Method
pc	3103.64	kPa	Joback Method
rinpol	1080.00		NIST Webbook
tb	463.00 ± 4.00	K	NIST Webbook
tb	466.00 ± 4.00	K	NIST Webbook
tb	471.50	K	NIST Webbook
tc	694.11	K	Joback Method
tf	245.05	K	Joback Method
vc	0.474	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	302.10	J/mol×K	657.42	Joback Method
cpg	291.74	J/mol×K	620.73	Joback Method
cpg	280.70	J/mol×K	584.04	Joback Method
cpg	268.93	J/mol×K	547.35	Joback Method
cpg	256.40	J/mol×K	510.66	Joback Method
cpg	243.09	J/mol×K	473.97	Joback Method
cpg	311.78	J/mol×K	694.11	Joback Method
dvisc	0.0033321	Paxs	245.05	Joback Method
dvisc	0.0002333	Paxs	473.97	Joback Method
dvisc	0.0002993	Paxs	435.82	Joback Method
dvisc	0.0004028	Paxs	397.66	Joback Method
dvisc	0.0005773	Paxs	359.51	Joback Method
dvisc	0.0009014	Paxs	321.36	Joback Method
dvisc	0.0015870	Paxs	283.20	Joback Method
hvapt	51.40	kJ/mol	389.50	NIST Webbook
hvapt	48.50	kJ/mol	440.50	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.41947e+01
Coeff. B	-3.82371e+03
Coeff. C	-7.22130e+01
Temperature range (K), min.	347.16
Temperature range (K), max.	502.66

## Sources

The Yaws Handbook of Vapor

Pressure:  
Crippen Method:

<https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure>

<http://pubs.acs.org/doi/abs/10.1021/ci9903071>

Crippen Method:

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

Joback Method:

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

McGowan Method:

<http://link.springer.com/article/10.1007/BF02311772>

NIST Webbook:

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C2621467&Units=SI>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfl:</b>	Liquid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpolar:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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