

# di-(1-Methyl-2-methoxybutyl)adipate

<b>Inchi:</b>	InChI=1S/C16H30O6/c1-11(19-5)13(3)21-15(17)9-7-8-10-16(18)22-14(4)12(2)20-6/h11-1
<b>InchiKey:</b>	CZSSMCJAVMVEBB-UHFFFAOYSA-N
<b>Formula:</b>	C16H30O6
<b>SMILES:</b>	COC(C)C(C)OC(=O)CCCCC(=O)OC(C)C(C)OC
<b>Mol. weight [g/mol]:</b>	318.41

## Physical Properties

Property code	Value	Unit	Source
gf	-603.76	kJ/mol	Joback Method
hf	-1148.73	kJ/mol	Joback Method
hfus	31.05	kJ/mol	Joback Method
hvap	72.79	kJ/mol	Joback Method
log10ws	-2.87		Crippen Method
logp	2.480		Crippen Method
mcvol	262.920	ml/mol	McGowan Method
pc	1409.07	kPa	Joback Method
rinpol	1790.00		NIST Webbook
tb	761.14	K	Joback Method
tc	946.78	K	Joback Method
tf	398.86	K	Joback Method
vc	0.992	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	795.69	J/molxK	761.14	Joback Method
cpg	869.60	J/molxK	915.84	Joback Method
cpg	856.85	J/molxK	884.90	Joback Method
cpg	843.08	J/molxK	853.96	Joback Method
cpg	828.29	J/molxK	823.02	Joback Method
cpg	812.49	J/molxK	792.08	Joback Method
cpg	881.30	J/molxK	946.78	Joback Method
dvisc	0.0000383	Paxs	761.14	Joback Method
dvisc	0.0000535	Paxs	700.76	Joback Method

dvisc	0.0000795	Paxs	640.38	Joback Method
dvisc	0.0001284	Paxs	580.00	Joback Method
dvisc	0.0002319	Paxs	519.62	Joback Method
dvisc	0.0004890	Paxs	459.24	Joback Method
dvisc	0.0012927	Paxs	398.86	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R541814&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R541814&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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