

# Perylene, 3-hexyl-

<b>Other names:</b>	3-n-Hexylperylene 3-Hexylperylene
<b>Inchi:</b>	InChI=1S/C26H24/c1-2-3-4-5-9-18-16-17-24-22-14-7-11-19-10-6-13-21(25(19)22)23-15-
<b>InchiKey:</b>	TZZMLMCAA VPEBH-UHFFFAOYSA-N
<b>Formula:</b>	C26H24
<b>SMILES:</b>	CCCCCc1ccc2c3cccc4cccc(c5cccc1c52)c43
<b>Mol. weight [g/mol]:</b>	336.47
<b>CAS:</b>	7350-91-6

## Physical Properties

Property code	Value	Unit	Source
gf	662.77	kJ/mol	Joback Method
hf	329.50	kJ/mol	Joback Method
hfus	46.63	kJ/mol	Joback Method
hvap	84.32	kJ/mol	Joback Method
log10ws	-10.58		Crippen Method
logp	7.860		Crippen Method
mvol	279.900	ml/mol	McGowan Method
pc	1539.08	kPa	Joback Method
tb	909.10	K	Joback Method
tc	1146.72	K	Joback Method
tf	414.65 ± 0.50	K	NIST Webbook
vc	1.101	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	856.26	J/mol×K	909.10	Joback Method
cpg	940.16	J/mol×K	1107.12	Joback Method
cpg	923.02	J/mol×K	1067.51	Joback Method
cpg	906.29	J/mol×K	1027.91	Joback Method
cpg	889.73	J/mol×K	988.31	Joback Method
cpg	873.13	J/mol×K	948.70	Joback Method
cpg	957.92	J/mol×K	1146.72	Joback Method

dvisc	0.0015978	Paxs	909.10	Joback Method
dvisc	0.0017159	Paxs	856.98	Joback Method
dvisc	0.0018598	Paxs	804.85	Joback Method
dvisc	0.0020384	Paxs	752.73	Joback Method
dvisc	0.0022648	Paxs	700.61	Joback Method
dvisc	0.0025594	Paxs	648.48	Joback Method
dvisc	0.0029547	Paxs	596.36	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C7350916&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C7350916&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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