

# Heptane, 4-propyl-

<b>Other names:</b>	4-Propylheptane
<b>Inchi:</b>	InChI=1S/C10H22/c1-4-7-10(8-5-2)9-6-3/h10H,4-9H2,1-3H3
<b>InchiKey:</b>	ABYGSZMCWVXFCQ-UHFFFAOYSA-N
<b>Formula:</b>	C10H22
<b>SMILES:</b>	CCCC(CCC)CCC
<b>Mol. weight [g/mol]:</b>	142.28
<b>CAS:</b>	3178-29-8

## Physical Properties

Property code	Value	Unit	Source
af	0.4440		KDB
ap	349.150	K	KDB
gf	30.88	kJ/mol	Joback Method
hcg	6778.75	kJ/mol	KDB
hcn	6294.661	kJ/mol	KDB
hf	-255.01	kJ/mol	Joback Method
hfus	18.13	kJ/mol	Joback Method
hvap	48.50	kJ/mol	NIST Webbook
log10ws	-3.77		Crippen Method
logp	4.003		Crippen Method
mcvol	151.760	ml/mol	McGowan Method
pc	2180.00	kPa	KDB
rinpol	945.00		NIST Webbook
rinpol	945.00		NIST Webbook
rinpol	945.00		NIST Webbook
rinpol	906.00		NIST Webbook
rinpol	945.00		NIST Webbook
rinpol	907.70		NIST Webbook
rinpol	946.20		NIST Webbook
rinpol	946.20		NIST Webbook
rinpol	940.00		NIST Webbook
rinpol	906.00		NIST Webbook
tb	430.70	K	KDB
tb	430.00 ± 4.00	K	NIST Webbook
tb	435.00 ± 6.00	K	NIST Webbook
tb	435.00 ± 0.60	K	NIST Webbook
tc	601.00	K	KDB

tf	219.00	K	KDB
vc	0.545	m <sup>3</sup> /kmol	KDB
zc	0.2377620		KDB

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	387.16	J/mol×K	566.07	Joback Method
cpg	360.07	J/mol×K	510.75	Joback Method
cpg	345.71	J/mol×K	483.08	Joback Method
cpg	330.79	J/mol×K	455.42	Joback Method
cpg	315.29	J/mol×K	427.76	Joback Method
cpg	373.88	J/mol×K	538.41	Joback Method
cpg	399.91	J/mol×K	593.73	Joback Method
dvisc	0.0108083	Paxs	187.46	Joback Method
dvisc	0.0002265	Paxs	427.76	Joback Method
dvisc	0.0003093	Paxs	387.71	Joback Method
dvisc	0.0004537	Paxs	347.66	Joback Method
dvisc	0.0007355	Paxs	307.61	Joback Method
dvisc	0.0013777	Paxs	267.56	Joback Method
dvisc	0.0032190	Paxs	227.51	Joback Method
hvapt	38.12	kJ/mol	430.70	KDB
hvapt	44.10	kJ/mol	380.50	NIST Webbook
rfi	1.41130		298.15	KDB

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.42938e+01
Coeff. B	-3.54982e+03
Coeff. C	-6.37720e+01
Temperature range (K), min.	317.22
Temperature range (K), max.	458.97

Information

Value

Property code	pvap
Equation	$\ln(P_{vp}) = A + B/T + C \cdot \ln(T) + D \cdot T^2$
Coeff. A	1.06242e+02
Coeff. B	-9.34666e+03
Coeff. C	-1.34186e+01
Coeff. D	7.90603e-06
Temperature range (K), min.	317.15
Temperature range (K), max.	601.00

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C3178298&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C3178298&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>KDB Vapor Pressure Data:</b>	<a href="https://www.chemic.org/research/kdb/hcprop/showprop.php?cmpid=116">https://www.chemic.org/research/kdb/hcprop/showprop.php?cmpid=116</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>KDB:</b>	<a href="https://www.chemic.org/files/research/kdb/mol/mol116.mol">https://www.chemic.org/files/research/kdb/mol/mol116.mol</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>af:</b>	Acentric Factor
<b>ap:</b>	Aniline Point
<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hcg:</b>	Heat of Combustion, Gross form
<b>hcn:</b>	Heat of Combustion, Net Form
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure

<b>rfi:</b>	Refractive Index
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume
<b>zc:</b>	Critical Compressibility

Latest version available from:

<https://www.cheméo.com/cid/15-106-7/Heptane-4-propyl.pdf>

Generated by Cheméo on 2024-04-19 19:19:55.751313175 +0000 UTC m=+15843644.671890487.

Cheméo (<https://www.cheméo.com>) is the biggest free database of chemical and physical data for the process industry.