

# Erbon

**Other names:**

Propanoic acid, 2,2-dichloro-, 2-(2,4,5-trichlorophenoxy)ethyl ester  
Propionic acid, 2,2-dichloro-, 2-(2,4,5-trichlorophenoxy)ethyl ester  
Baron  
Ethanol, 2-(2,4,5-trichlorophenoxy)-, 2,2-dichloropropionate  
Novon  
Pentanate  
2-(2,4,5-Trichlorophenoxy)ethyl 2,2-dichloropropionate  
2,2-Dichloropropionic acid 2-(2,4,5-trichlorophenoxy)ethyl ester  
ERBN  
Novege

**Inchi:**

InChI=1S/C11H9Cl5O3/c1-11(15,16)10(17)19-3-2-18-9-5-7(13)6(12)4-8(9)14/h4-5H,2-3H

**InchiKey:**

KMHZPJNVPCAUMN-UHFFFAOYSA-N

**Formula:**

C11H9Cl5O3

**SMILES:**

CC(Cl)(Cl)C(=O)OCCOc1cc(Cl)c(Cl)cc1Cl

**Mol. weight [g/mol]:**

366.45

**CAS:**

136-25-4

## Physical Properties

Property code	Value	Unit	Source
gf	-270.47	kJ/mol	Joback Method
hf	-532.72	kJ/mol	Joback Method
hfus	34.67	kJ/mol	Joback Method
hvap	76.54	kJ/mol	Joback Method
log10ws	-5.10		Crippen Method
logp	4.763		Crippen Method
mcvol	216.600	ml/mol	McGowan Method
pc	2214.53	kPa	Joback Method
tb	775.33	K	Joback Method
tc	1012.66	K	Joback Method
tf	524.12	K	Joback Method
vc	0.820	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	486.90	J/molxK	775.33	Joback Method
cpg	496.04	J/molxK	814.88	Joback Method
cpg	504.34	J/molxK	854.44	Joback Method
cpg	511.82	J/molxK	893.99	Joback Method
cpg	518.50	J/molxK	933.55	Joback Method
cpg	524.42	J/molxK	973.10	Joback Method
cpg	529.60	J/molxK	1012.66	Joback Method
dvisc	0.0004581	Paxs	524.12	Joback Method
dvisc	0.0003080	Paxs	565.99	Joback Method
dvisc	0.0002187	Paxs	607.86	Joback Method
dvisc	0.0001623	Paxs	649.72	Joback Method
dvisc	0.0001249	Paxs	691.59	Joback Method
dvisc	0.0000990	Paxs	733.46	Joback Method
dvisc	0.0000805	Paxs	775.33	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C136254&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C136254&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient

<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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