

# (trans-4,5-Methylene)-7-octenyl-cyclopropane

<b>Inchi:</b>	InChI=1S/C12H20/c1-2-4-11-9-12(11)6-3-5-10-7-8-10/h2,10-12H,1,3-9H2/t11-,12-/m0/s1
<b>InchiKey:</b>	QGVURVXQAOWBIY-RYUDHWPXSA-N
<b>Formula:</b>	C12H20
<b>SMILES:</b>	C=CCC1CC1CCCC1CC1
<b>Mol. weight [g/mol]:</b>	164.29

## Physical Properties

Property code	Value	Unit	Source
gf	251.79	kJ/mol	Joback Method
hf	-40.32	kJ/mol	Joback Method
hfus	22.90	kJ/mol	Joback Method
hvap	41.15	kJ/mol	Joback Method
log10ws	-3.76		Crippen Method
logp	3.779		Crippen Method
mcvol	153.920	ml/mol	McGowan Method
pc	2265.42	kPa	Joback Method
rinpol	1162.00		NIST Webbook
rinpol	1162.00		NIST Webbook
tb	479.45	K	Joback Method
tc	668.79	K	Joback Method
tf	254.88	K	Joback Method
vc	0.602	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	362.09	J/molxK	479.45	Joback Method
cpg	447.60	J/molxK	637.24	Joback Method
cpg	432.43	J/molxK	605.68	Joback Method
cpg	416.36	J/molxK	574.12	Joback Method
cpg	399.32	J/molxK	542.56	Joback Method
cpg	381.25	J/molxK	511.01	Joback Method
cpg	461.91	J/molxK	668.79	Joback Method
dvisc	0.0008100	Paxs	479.45	Joback Method

dvisc	0.0008238	Paxs	442.02	Joback Method
dvisc	0.0008405	Paxs	404.59	Joback Method
dvisc	0.0008611	Paxs	367.16	Joback Method
dvisc	0.0008869	Paxs	329.74	Joback Method
dvisc	0.0009206	Paxs	292.31	Joback Method
dvisc	0.0009660	Paxs	254.88	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R138131&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R138131&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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