

# Creatine

<b>Other names:</b>	(«alpha»-Methylguanido)acetic acid 6020-87-7 (hydrate) Creatin Glycine, N-(aminoiminomethyl)-N-methyl- Kreatin Krebiozon Methylguanidoacetic acid N-(Aminoiminomethyl)-N-methyl-glycine N-Methyl-N-guanylglycine N-amidinosarcasine N-amidinosarcosine N-guanyl-N-methylglycine NSC 8752
<b>Inchi:</b>	InChI=1S/C4H9N3O2/c1-7(4(5)6)2-3(8)9/h2H2,1H3,(H3,5,6)(H,8,9)
<b>InchiKey:</b>	CVSVTCORWBXHQV-UHFFFAOYSA-N
<b>Formula:</b>	C4H9N3O2
<b>SMILES:</b>	CN(CC(=O)O)C(=N)N
<b>Mol. weight [g/mol]:</b>	131.13
<b>CAS:</b>	57-00-1

## Physical Properties

Property code	Value	Unit	Source
chs	-2325.50	kJ/mol	NIST Webbook
chs	-2323.10 ± 0.84	kJ/mol	NIST Webbook
gf	97.89	kJ/mol	Joback Method
hf	-91.05	kJ/mol	Joback Method
hfs	-537.18 ± 0.88	kJ/mol	NIST Webbook
hvap	72.69	kJ/mol	Joback Method
log10ws	-0.51		Crippen Method
logp	-1.104		Crippen Method
mcvol	100.300	ml/mol	McGowan Method
ss	189.50	J/mol×K	NIST Webbook
tb	606.28	K	Joback Method
tf	430.10	K	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	252.05	J/mol×K	606.28	Joback Method
cpg	50.00	J/mol×K	100.12	Joback Method
cpg	50.00	J/mol×K	100.12	Joback Method
cpg	50.00	J/mol×K	100.12	Joback Method
cpg	50.00	J/mol×K	100.12	Joback Method
cpg	50.00	J/mol×K	100.12	Joback Method
cpg	50.00	J/mol×K	100.12	Joback Method
cps	184.50	J/mol×K	323.00	NIST Webbook
cps	171.10	J/mol×K	296.30	NIST Webbook

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C57001&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C57001&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>The organisation of water around creatine and creatinine molecules:</b>	<a href="https://www.doi.org/10.1016/j.jct.2018.08.007">https://www.doi.org/10.1016/j.jct.2018.08.007</a>

## Legend

<b>chs:</b>	Standard solid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>cps:</b>	Solid phase heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfs:</b>	Solid phase enthalpy of formation at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>ss:</b>	Solid phase molar entropy at standard conditions
<b>tb:</b>	Normal Boiling Point Temperature

**tf:** Normal melting (fusion) point

Latest version available from:

<https://www.cheméo.com/cid/15-170-6/Creatine.pdf>

Generated by Cheméo on 2024-04-24 20:38:53.43708616 +0000 UTC m=+16280382.357663493.

Cheméo (<https://www.cheméo.com>) is the biggest free database of chemical and physical data for the process industry.