

# Cyclooctane, methyl-

<b>Other names:</b>	Methylcyclooctane
<b>Inchi:</b>	InChI=1S/C9H18/c1-9-7-5-3-2-4-6-8-9/h9H,2-8H2,1H3
<b>InchiKey:</b>	POCNHGFJLGYFIK-UHFFFAOYSA-N
<b>Formula:</b>	C9H18
<b>SMILES:</b>	CC1CCCCCCC1
<b>Mol. weight [g/mol]:</b>	126.24
<b>CAS:</b>	1502-38-1

## Physical Properties

Property code	Value	Unit	Source
gf	25.15	kJ/mol	Joback Method
hf	-187.09	kJ/mol	Joback Method
hfus	6.70	kJ/mol	Joback Method
hvap	36.40	kJ/mol	Joback Method
ie	9.70 ± 0.05	eV	NIST Webbook
log10ws	-3.24		Crippen Method
logp	3.367		Crippen Method
mcvol	126.810	ml/mol	McGowan Method
pc	2995.87	kPa	Joback Method
rinpol	1009.00		NIST Webbook
rinpol	1019.00		NIST Webbook
rinpol	999.00		NIST Webbook
rinpol	995.00		NIST Webbook
rinpol	992.00		NIST Webbook
rinpol	995.00		NIST Webbook
tb	433.41	K	Joback Method
tc	649.15	K	Joback Method
tf	288.00 ± 5.00	K	NIST Webbook
vc	0.457	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	254.12	J/mol×K	433.41	Joback Method

cpg	274.97	J/molxK	469.37	Joback Method
cpg	294.81	J/molxK	505.32	Joback Method
cpg	313.65	J/molxK	541.28	Joback Method
cpg	331.52	J/molxK	577.24	Joback Method
cpg	348.41	J/molxK	613.19	Joback Method
cpg	364.35	J/molxK	649.15	Joback Method
dvisc	0.0303341	Paxs	191.53	Joback Method
dvisc	0.0064292	Paxs	231.84	Joback Method
dvisc	0.0021577	Paxs	272.16	Joback Method
dvisc	0.0009598	Paxs	312.47	Joback Method
dvisc	0.0005138	Paxs	352.78	Joback Method
dvisc	0.0003126	Paxs	393.10	Joback Method
dvisc	0.0002086	Paxs	433.41	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C1502381&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C1502381&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>KDB:</b>	<a href="https://www.cheric.org/files/research/kdb/mol/mol581.mol">https://www.cheric.org/files/research/kdb/mol/mol581.mol</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature

**tc:** Critical Temperature  
**tf:** Normal melting (fusion) point  
**vc:** Critical Volume

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