

# Benzo[c]phenanthrene

<b>Other names:</b>	3,4-Benzophenanthrene Benzo-3,4-phenanthrene Tetrahelcene 3,4-Benzphenanthrene
<b>Inchi:</b>	InChI=1S/C18H12/c1-3-7-16-13(5-1)9-11-15-12-10-14-6-2-4-8-17(14)18(15)16/h1-12H
<b>InchiKey:</b>	TUAHORSUHVUKBD-UHFFFAOYSA-N
<b>Formula:</b>	C18H12
<b>SMILES:</b>	<chem>c1ccc2c(c1)ccc1ccc3ccccc3c12</chem>
<b>Mol. weight [g/mol]:</b>	228.29
<b>CAS:</b>	195-19-7

## Physical Properties

Property code	Value	Unit	Source
chs	-8983.00 ± 1.90	kJ/mol	NIST Webbook
ea	0.55 ± 0.01	eV	NIST Webbook
ea	0.33	eV	NIST Webbook
gf	513.78	kJ/mol	Joback Method
hf	291.20	kJ/mol	NIST Webbook
hfs	184.90 ± 3.00	kJ/mol	NIST Webbook
hfus	26.70	kJ/mol	Joback Method
hsub	106.30	kJ/mol	NIST Webbook
hsub	106.00	kJ/mol	NIST Webbook
hvap	64.18	kJ/mol	Joback Method
ie	7.62	eV	NIST Webbook
ie	7.76	eV	NIST Webbook
ie	7.86	eV	NIST Webbook
ie	7.60	eV	NIST Webbook
ie	7.60 ± 0.02	eV	NIST Webbook
log10ws	-6.88		Crippen Method
logp	5.146		Crippen Method
mvol	182.340	ml/mol	McGowan Method
pc	2741.15	kPa	Joback Method
rinpol	2390.00		NIST Webbook
rinpol	391.24		NIST Webbook
rinpol	391.39		NIST Webbook
rinpol	390.90		NIST Webbook
rinpol	2400.20		NIST Webbook

rnpol	390.57		NIST Webbook
rnpol	391.07		NIST Webbook
rnpol	391.24		NIST Webbook
rnpol	2427.00		NIST Webbook
rnpol	391.10		NIST Webbook
rnpol	389.52		NIST Webbook
rnpol	388.58		NIST Webbook
rnpol	389.97		NIST Webbook
rnpol	390.27		NIST Webbook
rnpol	391.39		NIST Webbook
rnpol	2400.20		NIST Webbook
rnpol	2390.00		NIST Webbook
rnpol	391.07		NIST Webbook
rnpol	394.30		NIST Webbook
rnpol	391.12		NIST Webbook
rnpol	390.89		NIST Webbook
rnpol	391.07		NIST Webbook
rnpol	391.11		NIST Webbook
rnpol	391.30		NIST Webbook
rnpol	2400.00		NIST Webbook
rnpol	2390.00		NIST Webbook
rnpol	2400.00		NIST Webbook
rnpol	390.57		NIST Webbook
rnpol	389.36		NIST Webbook
rnpol	2427.00		NIST Webbook
tb	704.82	K	Joback Method
tc	968.39	K	Joback Method
tf	342.00 ± 1.00	K	NIST Webbook
tf	334.80 ± 0.80	K	NIST Webbook
tf	341.00 ± 1.00	K	NIST Webbook
vc	0.702	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	468.00	J/mol×K	704.82	Joback Method
cpg	482.86	J/mol×K	748.75	Joback Method
cpg	496.51	J/mol×K	792.68	Joback Method
cpg	509.15	J/mol×K	836.61	Joback Method
cpg	521.02	J/mol×K	880.54	Joback Method
cpg	532.31	J/mol×K	924.47	Joback Method

cpg	543.24	J/molxK	968.39	Joback Method
dvisc	0.0011669	Paxs	529.73	Joback Method
dvisc	0.0013935	Paxs	485.95	Joback Method
dvisc	0.0017235	Paxs	442.18	Joback Method
dvisc	0.0010040	Paxs	573.50	Joback Method
dvisc	0.0008825	Paxs	617.27	Joback Method
dvisc	0.0007890	Paxs	661.05	Joback Method
dvisc	0.0007153	Paxs	704.82	Joback Method
hfust	15.50	kJ/mol	339.20	NIST Webbook
hfust	16.32	kJ/mol	334.70	NIST Webbook
hfust	16.32	kJ/mol	334.70	NIST Webbook
hsubt	106.00	kJ/mol	293.00	NIST Webbook

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C195197&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C195197&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>chs:</b>	Standard solid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>ea:</b>	Electron affinity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfs:</b>	Solid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hfust:</b>	Enthalpy of fusion at a given temperature
<b>hsub:</b>	Enthalpy of sublimation at standard conditions
<b>hsubt:</b>	Enthalpy of sublimation at a given temperature
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient

<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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