

# 10-Methyl-2-undecanol

<b>Inchi:</b>	InChI=1S/C12H26O/c1-11(2)9-7-5-4-6-8-10-12(3)13/h11-13H,4-10H2,1-3H3
<b>InchiKey:</b>	QQZWBFBYOACMAO-UHFFFAOYSA-N
<b>Formula:</b>	C12H26O
<b>SMILES:</b>	CC(C)CCCCCCC(C)O
<b>Mol. weight [g/mol]:</b>	186.33

## Physical Properties

Property code	Value	Unit	Source
gf	-91.54	kJ/mol	Joback Method
hf	-453.80	kJ/mol	Joback Method
hfus	23.88	kJ/mol	Joback Method
hvap	58.21	kJ/mol	Joback Method
log10ws	-3.98		Crippen Method
logp	3.754		Crippen Method
mcvol	185.810	ml/mol	McGowan Method
pc	1954.41	kPa	Joback Method
rinpol	1373.00		NIST Webbook
rinpol	1373.00		NIST Webbook
tb	565.26	K	Joback Method
tc	728.05	K	Joback Method
tf	255.82	K	Joback Method
vc	0.715	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	479.00	J/molxK	565.26	Joback Method
cpg	494.19	J/molxK	592.39	Joback Method
cpg	508.77	J/molxK	619.52	Joback Method
cpg	522.76	J/molxK	646.65	Joback Method
cpg	536.16	J/molxK	673.79	Joback Method
cpg	549.01	J/molxK	700.92	Joback Method
cpg	561.31	J/molxK	728.05	Joback Method
dvisc	0.0773422	Paxs	255.82	Joback Method

dvisc	0.0092648	Paxs	307.39	Joback Method
dvisc	0.0020420	Paxs	358.97	Joback Method
dvisc	0.0006581	Paxs	410.54	Joback Method
dvisc	0.0002731	Paxs	462.11	Joback Method
dvisc	0.0001352	Paxs	513.69	Joback Method
dvisc	0.0000761	Paxs	565.26	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R508768&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R508768&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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