

# 3,5,6-Trichloro-2,2,3,4,4,5,6,6-octafluorohexanoic acid

Inchi:  
acid

InChI=1S/C6HCl3F8O2/c7-3(12,2(10,11)1(18)19)5(14,15)4(8,13)6(9,16)17/h(H,18,19)

InchiKey:

ANRGSWRBGXZQLY-UHFFFAOYSA-N

Formula:

C6HCl3F8O2

SMILES:

O=C(O)C(F)(F)C(F)(Cl)C(F)(F)C(F)(Cl)C(F)(F)Cl

Mol. weight [g/mol]:

363.42

CAS:

2106-54-9

## Physical Properties

Property code	Value	Unit	Source
gf	-1846.17	kJ/mol	Joback Method
hf	-2091.83	kJ/mol	Joback Method
hfus	17.14	kJ/mol	Joback Method
hvap	52.51	kJ/mol	Joback Method
log10ws	-4.25		Crippen Method
logp	3.982		Crippen Method
mcvol	153.720	ml/mol	McGowan Method
pc	2462.92	kPa	Joback Method
tb	573.03	K	Joback Method
tc	744.30	K	Joback Method
tf	374.71	K	Joback Method
vc	0.632	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	377.76	J/molxK	573.03	Joback Method
cpg	384.30	J/molxK	601.58	Joback Method
cpg	390.05	J/molxK	630.12	Joback Method
cpg	395.10	J/molxK	658.67	Joback Method
cpg	399.51	J/molxK	687.21	Joback Method
cpg	403.34	J/molxK	715.76	Joback Method
cpg	406.67	J/molxK	744.30	Joback Method

# Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.99042e+01
Coeff. B	-7.72168e+03
Temperature range (K), min.	373.00
Temperature range (K), max.	529.14

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=B6003897&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=B6003897&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.cheméo.com/cid/15-339-9/3-5-6-Trichloro-2-2-3-4-4-5-6-6-octafluorohexanoic-acid.pdf>

Generated by Cheméo on 2024-04-27 15:37:22.352385776 +0000 UTC m=+16521491.272963089.

Cheméo (<https://www.cheméo.com>) is the biggest free database of chemical and physical data for the process industry.