

# Succinic acid, monochloride pent-4-en-2-yl ester

Inchi:	InChI=1S/C9H13ClO3/c1-3-4-7(2)13-9(12)6-5-8(10)11/h3,7H,1,4-6H2,2H3
InchiKey:	LBRRWHKGDPAWTB-UHFFFAOYSA-N
Formula:	C9H13ClO3
SMILES:	C=CCC(C)OC(=O)CCC(=O)Cl
Mol. weight [g/mol]:	204.65

## Physical Properties

Property code	Value	Unit	Source
gf	-264.47	kJ/mol	Joback Method
hf	-482.06	kJ/mol	Joback Method
hfus	22.85	kJ/mol	Joback Method
hvap	54.86	kJ/mol	Joback Method
log10ws	-2.35		Crippen Method
logp	2.040		Crippen Method
mcvol	154.620	ml/mol	McGowan Method
pc	2603.08	kPa	Joback Method
rinpol	1275.00		NIST Webbook
rinpol	1275.00		NIST Webbook
tb	569.15	K	Joback Method
tc	763.87	K	Joback Method
tf	326.44	K	Joback Method
vc	0.594	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	351.08	J/molxK	569.15	Joback Method
cpg	362.66	J/molxK	601.60	Joback Method
cpg	373.65	J/molxK	634.06	Joback Method
cpg	384.07	J/molxK	666.51	Joback Method
cpg	393.93	J/molxK	698.96	Joback Method
cpg	403.24	J/molxK	731.42	Joback Method
cpg	412.01	J/molxK	763.87	Joback Method
dvisc	0.0029667	Paxs	326.44	Joback Method

dvisc	0.0015530	Paxs	366.89	Joback Method
dvisc	0.0009245	Paxs	407.34	Joback Method
dvisc	0.0006044	Paxs	447.79	Joback Method
dvisc	0.0004240	Paxs	488.25	Joback Method
dvisc	0.0003140	Paxs	528.70	Joback Method
dvisc	0.0002427	Paxs	569.15	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U353459&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U353459&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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