

# Cyclohexene, 1-nonyl-

<b>Other names:</b>	1-Nonyl-1-cyclohexene
<b>Inchi:</b>	InChI=1S/C15H28/c1-2-3-4-5-6-7-9-12-15-13-10-8-11-14-15/h13H,2-12,14H2,1H3
<b>InchiKey:</b>	GTEZIRITHHEDAC-UHFFFAOYSA-N
<b>Formula:</b>	C15H28
<b>SMILES:</b>	CCCCCCCCC1=CCCCC1
<b>Mol. weight [g/mol]:</b>	208.38
<b>CAS:</b>	15232-88-9

## Physical Properties

Property code	Value	Unit	Source
gf	127.91	kJ/mol	Joback Method
hf	-231.96	kJ/mol	Joback Method
hfus	26.20	kJ/mol	Joback Method
hvap	50.68	kJ/mol	Joback Method
log10ws	-5.85		Crippen Method
logp	5.627		Crippen Method
mcvol	207.050	ml/mol	McGowan Method
pc	1737.56	kPa	Joback Method
ripol	1680.00		NIST Webbook
ripol	1712.00		NIST Webbook
ripol	1716.00		NIST Webbook
ripol	1665.00		NIST Webbook
ripol	1685.00		NIST Webbook
ripol	1701.00		NIST Webbook
ripol	1697.00		NIST Webbook
ripol	1702.00		NIST Webbook
ripol	1680.00		NIST Webbook
ripol	1690.90		NIST Webbook
ripol	1696.90		NIST Webbook
ripol	1702.00		NIST Webbook
ripol	1685.00		NIST Webbook
ripol	1691.00		NIST Webbook
tb	570.96	K	Joback Method
tc	759.32	K	Joback Method
tf	283.71	K	Joback Method
vc	0.795	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	532.09	J/molxK	570.96	Joback Method
cpg	552.70	J/molxK	602.35	Joback Method
cpg	572.29	J/molxK	633.75	Joback Method
cpg	590.90	J/molxK	665.14	Joback Method
cpg	608.56	J/molxK	696.53	Joback Method
cpg	625.30	J/molxK	727.93	Joback Method
cpg	641.17	J/molxK	759.32	Joback Method
dvisc	0.0050407	Paxs	283.71	Joback Method
dvisc	0.0018707	Paxs	331.59	Joback Method
dvisc	0.0008915	Paxs	379.46	Joback Method
dvisc	0.0005016	Paxs	427.34	Joback Method
dvisc	0.0003169	Paxs	475.21	Joback Method
dvisc	0.0002178	Paxs	523.09	Joback Method
dvisc	0.0001594	Paxs	570.96	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C15232889&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C15232889&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions

<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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