

2-Chlorobutyramide

Inchi:	InChI=1S/C4H8ClNO/c1-2-3(5)4(6)7/h3H,2H2,1H3,(H2,6,7)
InchiKey:	VJIOSCMTZVXQQU-UHFFFAOYSA-N
Formula:	C4H8ClNO
SMILES:	CCC(Cl)C(N)=O
Mol. weight [g/mol]:	121.56
CAS:	7462-73-9

Physical Properties

Property code	Value	Unit	Source
gf	-94.04	kJ/mol	Joback Method
hf	-225.70	kJ/mol	Joback Method
hfus	13.59	kJ/mol	Joback Method
hvap	45.88	kJ/mol	Joback Method
log10ws	-0.97		Crippen Method
logp	0.489		Crippen Method
mvol	91.010	ml/mol	McGowan Method
pc	4379.97	kPa	Joback Method
tb	454.31	K	Joback Method
tc	661.30	K	Joback Method
tf	282.95	K	Joback Method
vc	0.338	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	174.28	J/mol×K	454.31	Joback Method
cpg	182.33	J/mol×K	488.81	Joback Method
cpg	189.97	J/mol×K	523.31	Joback Method
cpg	197.21	J/mol×K	557.81	Joback Method
cpg	204.05	J/mol×K	592.31	Joback Method
cpg	210.52	J/mol×K	626.81	Joback Method
cpg	216.63	J/mol×K	661.30	Joback Method

Sources

NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C7462739&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mccvol:	McGowan's characteristic volume
pc:	Critical Pressure
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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