

# Benzene, 3-cyclohexen-1-yl-

<b>Other names:</b>	4-Phenylcyclohexene
<b>Inchi:</b>	InChI=1S/C12H14/c1-3-7-11(8-4-1)12-9-5-2-6-10-12/h1-5,7-8,12H,6,9-10H2
<b>InchiKey:</b>	XWCWNUSFQVJNDI-UHFFFAOYSA-N
<b>Formula:</b>	C12H14
<b>SMILES:</b>	C1=CCC(c2ccccc2)CC1
<b>Mol. weight [g/mol]:</b>	158.24
<b>CAS:</b>	4994-16-5

## Physical Properties

Property code	Value	Unit	Source
gf	216.98	kJ/mol	Joback Method
hf	57.62	kJ/mol	Joback Method
hfus	13.93	kJ/mol	Joback Method
hvap	45.30	kJ/mol	Joback Method
ie	8.57 ± 0.01	eV	NIST Webbook
log10ws	-3.66		Crippen Method
logp	3.510		Crippen Method
mcvol	141.020	ml/mol	McGowan Method
pc	3093.29	kPa	Joback Method
rinpol	1345.00		NIST Webbook
rinpol	1345.00		NIST Webbook
ripol	1800.00		NIST Webbook
ripol	1802.00		NIST Webbook
ripol	1800.00		NIST Webbook
tb	519.35	K	Joback Method
tc	762.70	K	Joback Method
tf	259.56	K	Joback Method
vc	0.518	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	315.09	J/mol×K	519.35	Joback Method
cpg	335.13	J/mol×K	559.91	Joback Method

cpg	353.72	J/mol×K	600.47	Joback Method
cpg	370.92	J/mol×K	641.02	Joback Method
cpg	386.81	J/mol×K	681.58	Joback Method
cpg	401.46	J/mol×K	722.14	Joback Method
cpg	414.92	J/mol×K	762.70	Joback Method
dvisc	0.0041071	Paxs	259.56	Joback Method
dvisc	0.0017955	Paxs	302.86	Joback Method
dvisc	0.0009654	Paxs	346.16	Joback Method
dvisc	0.0005959	Paxs	389.46	Joback Method
dvisc	0.0004051	Paxs	432.75	Joback Method
dvisc	0.0002954	Paxs	476.05	Joback Method
dvisc	0.0002271	Paxs	519.35	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C4994165&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C4994165&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307I">http://pubs.acs.org/doi/abs/10.1021/ci990307I</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

**tf:** Normal melting (fusion) point

**vc:** Critical Volume

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