

# 2-Propyn-1-amine, N,N-diethyl-

<b>Other names:</b>	2-Propynylamine, N,N-diethyl- Diethylpropargylamine N,N-Diethyl-2-propynylamine N,N-Diethylpropargylamine 1-N,N-Diethylamino-2-propyne 3-(Diethylamino)propyne 3-(N,N-Diethylamino)propyne 1-Diethylaminopropyne-2 3-Diethylamino-1-propyne 1-Diethylamino-2-propyne NSC 63868
<b>Inchi:</b>	InChI=1S/C7H13N/c1-4-7-8(5-2)6-3/h1H,5-7H2,2-3H3
<b>InchiKey:</b>	JZJXKEWVUBVOEH-UHFFFAOYSA-N
<b>Formula:</b>	C7H13N
<b>SMILES:</b>	C#CCN(CC)CC
<b>Mol. weight [g/mol]:</b>	111.18
<b>CAS:</b>	4079-68-9

## Physical Properties

Property code	Value	Unit	Source
gf	341.91	kJ/mol	Joback Method
hf	171.62	kJ/mol	Joback Method
hfus	19.88	kJ/mol	Joback Method
hvap	33.08	kJ/mol	Joback Method
log10ws	-1.12		Crippen Method
logp	0.961		Crippen Method
mcvol	110.870	ml/mol	McGowan Method
pc	3287.81	kPa	Joback Method
rinpol	757.00		NIST Webbook
rinpol	757.00		NIST Webbook
tb	362.12	K	Joback Method
tc	536.13	K	Joback Method
tf	248.09	K	Joback Method
vc	0.407	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	197.70	J/mol×K	362.12	Joback Method
cpg	209.38	J/mol×K	391.12	Joback Method
cpg	220.53	J/mol×K	420.12	Joback Method
cpg	231.16	J/mol×K	449.13	Joback Method
cpg	241.30	J/mol×K	478.13	Joback Method
cpg	250.97	J/mol×K	507.13	Joback Method
cpg	260.18	J/mol×K	536.13	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C4079689&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C4079689&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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