

# Flucythrinate

**Other names:**

AC 222705

Benzeneacetic acid, 4-(difluoromethoxy)-«alpha»-(1-methylethyl)-,  
cyano(3-phenoxyphenyl)methyl ester  
Benzeneacetic acid, 4-(difluoromethoxy)-«alpha»-(1-methylethyl)-,  
cyano(3-phenoxyphenyl)methyl ester  
CyBolt

Flucythrinate, isomer 1

Flucythrinate, isomer 2

Fluorocythrin

Pay-Off

[cyano-[3-(phenoxy)phenyl]methyl]

2-[4-(difluoromethoxy)phenyl]-3-methylbutanoate

cyano(3-phenoxyphenyl)methyl 2-[4-(difluoromethoxy)phenyl]-3-methylbutyrate

**Inchi:**

InChI=1S/C26H23F2NO4/c1-17(2)24(18-11-13-21(14-12-18)32-26(27)28)25(30)33-23(16)

**InchiKey:**

GBIHOLCMZGAKNG-UHFFFAOYSA-N

**Formula:**

C26H23F2NO4

**SMILES:**

CC(C)C(C(=O)OC(C#N)c1cccc(Oc2ccccc2)c1)c1ccc(OC(F)F)cc1

**Mol. weight [g/mol]:**

451.46

**CAS:**

70124-77-5

## Physical Properties

Property code	Value	Unit	Source
gf	-224.11	kJ/mol	Joback Method
hf	-651.02	kJ/mol	Joback Method
hfus	43.18	kJ/mol	Joback Method
hvap	102.89	kJ/mol	Joback Method
log10ws	-6.88		Aqueous Solubility Prediction Method
log10ws	-6.88		Estimated Solubility Method
logp	6.628		Crippen Method
mcvol	330.020	ml/mol	McGowan Method
pc	1255.70	kPa	Joback Method
rinpol	2871.00		NIST Webbook
rinpol	2874.00		NIST Webbook
rinpol	2844.00		NIST Webbook
rinpol	2844.00		NIST Webbook
rinpol	2847.00		NIST Webbook
tb	1104.27	K	Joback Method
tc	1356.15	K	Joback Method

tf	609.87	K	Joback Method
vc	1.266	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1063.85	J/mol×K	1104.27	Joback Method
cpg	1071.79	J/mol×K	1146.25	Joback Method
cpg	1078.03	J/mol×K	1188.23	Joback Method
cpg	1082.64	J/mol×K	1230.21	Joback Method
cpg	1085.69	J/mol×K	1272.19	Joback Method
cpg	1087.24	J/mol×K	1314.17	Joback Method
cpg	1087.37	J/mol×K	1356.15	Joback Method

## Sources

<b>Estimated Solubility Method:</b>	<a href="http://pubs.acs.org/doi/suppl/10.1021/ci034243x/suppl_file/ci034243xsi20040112_053635.txt">http://pubs.acs.org/doi/suppl/10.1021/ci034243x/suppl_file/ci034243xsi20040112_053635.txt</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C70124775&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C70124775&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>Aqueous Solubility Prediction Method:</b>	<a href="http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDa">http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDa</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices

**tb:** Normal Boiling Point Temperature  
**tc:** Critical Temperature  
**tf:** Normal melting (fusion) point  
**vc:** Critical Volume

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