

# Benzenemethanol, 3-nitro-

<b>Other names:</b>	Benzyl alcohol, m-nitro- m-Nitrobenzyl alcohol 3-Nitrobenzyl alcohol (3-Nitrophenyl) methanol
<b>Inchi:</b>	InChI=1S/C7H7NO3/c9-5-6-2-1-3-7(4-6)8(10)11/h1-4,9H,5H2
<b>InchiKey:</b>	CWNPOQFCIIFQDM-UHFFFAOYSA-N
<b>Formula:</b>	C7H7NO3
<b>SMILES:</b>	O=[N+](O)c1cccc(CO)c1
<b>Mol. weight [g/mol]:</b>	153.14
<b>CAS:</b>	619-25-0

## Physical Properties

Property code	Value	Unit	Source
gf	9.57	kJ/mol	Joback Method
hf	-125.74	kJ/mol	Joback Method
hfus	22.99	kJ/mol	Joback Method
hvap	67.38	kJ/mol	Joback Method
log10ws	-2.27		Crippen Method
logp	1.087		Crippen Method
mcvol	109.020	ml/mol	McGowan Method
pc	4627.70	kPa	Joback Method
rinpol	1545.00		NIST Webbook
tb	635.24	K	Joback Method
tc	863.19	K	Joback Method
tf	412.02	K	Joback Method
vc	0.420	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	262.07	J/molxK	635.24	Joback Method
cpg	270.91	J/molxK	673.23	Joback Method
cpg	279.10	J/molxK	711.22	Joback Method
cpg	286.68	J/molxK	749.21	Joback Method

cpg	293.68	J/mol×K	787.21	Joback Method
cpg	300.14	J/mol×K	825.20	Joback Method
cpg	306.09	J/mol×K	863.19	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	450.70	K	0.40	NIST Webbook

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C619250&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C619250&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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