

# 1-Carboxycyclopropane-2-acetic acid (Z), dimethyl ester

<b>Other names:</b>	Methyl 2-(2-methoxy-2-oxoethyl)cyclopropanecarboxylate, (Z)- 2-Methoxycarbonylmethyl-cyclopropanecarboxylic acid methyl ester, Z
<b>Inchi:</b>	InChI=1S/C8H12O4/c1-11-7(9)4-5-3-6(5)8(10)12-2/h5-6H,3-4H2,1-2H3/t5-,6-/m1/s1
<b>InchiKey:</b>	UXTUIDNFTDLRIX-PHDIDXHHSA-N
<b>Formula:</b>	C8H12O4
<b>SMILES:</b>	COC(=O)CC1CC1C(=O)OC
<b>Mol. weight [g/mol]:</b>	172.18
<b>CAS:</b>	77462-53-4

## Physical Properties

Property code	Value	Unit	Source
gf	-398.32	kJ/mol	Joback Method
hf	-645.59	kJ/mol	Joback Method
hfus	21.26	kJ/mol	Joback Method
hvap	51.32	kJ/mol	Joback Method
log10ws	-0.31		Crippen Method
logp	0.359		Crippen Method
mcvol	127.600	ml/mol	McGowan Method
pc	3096.73	kPa	Joback Method
rinp	1175.00		NIST Webbook
rinp	1175.00		NIST Webbook
tb	537.09	K	Joback Method
tc	733.84	K	Joback Method
tf	337.94	K	Joback Method
vc	0.487	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	308.99	J/mol×K	537.09	Joback Method
cpg	321.47	J/mol×K	569.88	Joback Method
cpg	333.36	J/mol×K	602.67	Joback Method
cpg	344.67	J/mol×K	635.46	Joback Method
cpg	355.42	J/mol×K	668.25	Joback Method

cpg	365.61	J/molxK	701.05	Joback Method
cpg	375.24	J/molxK	733.84	Joback Method
dvisc	0.0016772	Paxs	337.94	Joback Method
dvisc	0.0012748	Paxs	371.13	Joback Method
dvisc	0.0010135	Paxs	404.32	Joback Method
dvisc	0.0008344	Paxs	437.51	Joback Method
dvisc	0.0007060	Paxs	470.71	Joback Method
dvisc	0.0006106	Paxs	503.90	Joback Method
dvisc	0.0005377	Paxs	537.09	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C77462534&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C77462534&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvac:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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