

# Benzene, 1,1'-(1,2-dibromo-1,2-ethanediyl)bis-

<b>Other names:</b>	1,2-Dibromo-1,2-diphenylethane Stilbene dibromide «alpha», «alpha»-Dibromobiphenyl Biphenyl, «alpha», «alpha»'-dibromo- (1,2-Dibromo-2-phenylethyl)benzene 1,1'-(1,2-Dibromo-1,2-ethanediyl)bisbenzene
<b>Inchi:</b>	InChI=1S/C14H12Br2/c15-13(11-7-3-1-4-8-11)14(16)12-9-5-2-6-10-12/h1-10,13-14H
<b>InchiKey:</b>	GKESIQQTGWVOLH-UHFFFAOYSA-N
<b>Formula:</b>	C14H12Br2
<b>SMILES:</b>	BrC(c1ccccc1)C(Br)c1ccccc1
<b>Mol. weight [g/mol]:</b>	340.05
<b>CAS:</b>	5789-30-0

## Physical Properties

Property code	Value	Unit	Source
gf	315.58	kJ/mol	Joback Method
hf	182.87	kJ/mol	Joback Method
hfus	23.62	kJ/mol	Joback Method
hvap	63.40	kJ/mol	Joback Method
log10ws	-5.67		Crippen Method
logp	5.259		Crippen Method
mccvol	195.600	ml/mol	McGowan Method
pc	3280.28	kPa	Joback Method
tb	704.52	K	Joback Method
tc	979.66	K	Joback Method
tf	511.00 ± 4.00	K	NIST Webbook
vc	0.716	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	450.47	J/mol×K	704.52	Joback Method
cpg	465.02	J/mol×K	750.38	Joback Method
cpg	478.17	J/mol×K	796.23	Joback Method

cpg	490.08	J/molxK	842.09	Joback Method
cpg	500.91	J/molxK	887.94	Joback Method
cpg	510.80	J/molxK	933.80	Joback Method
cpg	519.93	J/molxK	979.66	Joback Method
dvisc	0.0018946	Paxs	389.98	Joback Method
dvisc	0.0009303	Paxs	442.40	Joback Method
dvisc	0.0005311	Paxs	494.83	Joback Method
dvisc	0.0003376	Paxs	547.25	Joback Method
dvisc	0.0002323	Paxs	599.67	Joback Method
dvisc	0.0001697	Paxs	652.10	Joback Method
dvisc	0.0001299	Paxs	704.52	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C5789300&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C5789300&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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