

# Propane, 1-(2-chloroethoxy)-

<b>Other names:</b>	2-Propoxyethyl chloride 1-(2-chloroethoxy)propane
<b>Inchi:</b>	InChI=1S/C5H11ClO/c1-2-4-7-5-3-6/h2-5H2,1H3
<b>InchiKey:</b>	BHDSGQOSIWVMJW-UHFFFAOYSA-N
<b>Formula:</b>	C5H11ClO
<b>SMILES:</b>	CCCOCCCI
<b>Mol. weight [g/mol]:</b>	122.59
<b>CAS:</b>	42149-74-6

## Physical Properties

Property code	Value	Unit	Source
gf	-125.71	kJ/mol	Joback Method
hf	-294.49	kJ/mol	Joback Method
hfus	14.09	kJ/mol	Joback Method
hvap	33.52	kJ/mol	Joback Method
log10ws	-1.16		Crippen Method
logp	1.652		Crippen Method
mvol	99.420	ml/mol	McGowan Method
pc	3257.86	kPa	Joback Method
tb	373.65	K	Joback Method
tc	546.90	K	Joback Method
tf	198.26	K	Joback Method
vc	0.383	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	172.70	J/mol×K	373.65	Joback Method
cpg	213.61	J/mol×K	518.02	Joback Method
cpg	205.94	J/mol×K	489.15	Joback Method
cpg	198.02	J/mol×K	460.27	Joback Method
cpg	189.84	J/mol×K	431.40	Joback Method
cpg	181.40	J/mol×K	402.52	Joback Method
cpg	221.03	J/mol×K	546.90	Joback Method

dvisc	0.0002660	Paxs	373.65	Joback Method
dvisc	0.0003395	Paxs	344.42	Joback Method
dvisc	0.0004535	Paxs	315.19	Joback Method
dvisc	0.0006426	Paxs	285.95	Joback Method
dvisc	0.0009858	Paxs	256.72	Joback Method
dvisc	0.0016880	Paxs	227.49	Joback Method
dvisc	0.0033874	Paxs	198.26	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C42149746&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C42149746&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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