

Pentane, 1-propoxy-

Other names:	1-propoxypentane 4-oxanonane ether, pentyl propyl pentyl propyl ether propyl pentyl ether
Inchi:	InChI=1S/C8H18O/c1-3-5-6-8-9-7-4-2/h3-8H2,1-2H3
InchiKey:	DMUSSSYUUUYJRM-UHFFFAOYSA-N
Formula:	C8H18O
SMILES:	CCCCCOCCC
Mol. weight [g/mol]:	130.23
CAS:	18641-82-2

Physical Properties

Property code	Value	Unit	Source
gf	-88.52	kJ/mol	Joback Method
hf	-340.67	kJ/mol	Joback Method
hfus	17.66	kJ/mol	Joback Method
hvap	42.81	kJ/mol	NIST Webbook
log10ws	-2.26		Crippen Method
logp	2.603		Crippen Method
mcvol	129.450	ml/mol	McGowan Method
pc	2492.52	kPa	Joback Method
tb	405.20	K	NIST Webbook
tc	568.81	K	Joback Method
tf	202.15	K	Joback Method
vc	0.501	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	305.91	J/molxK	514.16	Joback Method
cpg	282.47	J/molxK	459.51	Joback Method
cpg	270.17	J/molxK	432.19	Joback Method
cpg	257.48	J/molxK	404.86	Joback Method

cpg	317.06	J/molxK	541.49	Joback Method
cpg	327.83	J/molxK	568.81	Joback Method
cpg	294.38	J/molxK	486.84	Joback Method
dvisc	0.0017915	Paxs	235.94	Joback Method
dvisc	0.0009546	Paxs	269.72	Joback Method
dvisc	0.0005852	Paxs	303.50	Joback Method
dvisc	0.0003956	Paxs	337.29	Joback Method
dvisc	0.0002873	Paxs	371.08	Joback Method
dvisc	0.0041498	Paxs	202.15	Joback Method
dvisc	0.0002200	Paxs	404.86	Joback Method
hvapt	35.00	kJ/mol	405.20	NIST Webbook
rhoI	742.73	kg/m3	323.15	Excess Enthalpies in Binary Systems of Isomeric C8 Aliphatic Monoethers with Acetonitrile and Their Description by the COSMO-SAC Model
rhoI	747.19	kg/m3	318.15	Excess Enthalpies in Binary Systems of Isomeric C8 Aliphatic Monoethers with Acetonitrile and Their Description by the COSMO-SAC Model
rhoI	756.01	kg/m3	308.15	Excess Enthalpies in Binary Systems of Isomeric C8 Aliphatic Monoethers with Acetonitrile and Their Description by the COSMO-SAC Model
rhoI	760.40	kg/m3	303.15	Excess Enthalpies in Binary Systems of Isomeric C8 Aliphatic Monoethers with Acetonitrile and Their Description by the COSMO-SAC Model

rho1	764.78	kg/m3	298.15	Excess Enthalpies in Binary Systems of Isomeric C8 Aliphatic Monoethers with Acetonitrile and Their Description by the COSMO-SAC Model
rho1	769.12	kg/m3	293.15	Excess Enthalpies in Binary Systems of Isomeric C8 Aliphatic Monoethers with Acetonitrile and Their Description by the COSMO-SAC Model
rho1	773.45	kg/m3	288.15	Excess Enthalpies in Binary Systems of Isomeric C8 Aliphatic Monoethers with Acetonitrile and Their Description by the COSMO-SAC Model
rho1	751.60	kg/m3	313.15	Excess Enthalpies in Binary Systems of Isomeric C8 Aliphatic Monoethers with Acetonitrile and Their Description by the COSMO-SAC Model

Sources

Crippen Method:
Liquid-liquid equilibrium in binary systems of isomeric C8 aliphatic monoethers with acetonitrile.
Liquid-liquid equilibrium in binary systems of isomeric C8 aliphatic monoethers with acetonitrile.
Excess Enthalpies in Binary Systems of Isomeric C8 Aliphatic Monoethers with Acetonitrile and Their Description by the COSMO-SAC Model: McGowan Method:
NIST Webbook:
Crippen Method:

https://www.chemeo.com/doc/models/crippen_log10ws
<https://www.doi.org/10.1016/j.fluid.2013.07.002>
<https://www.doi.org/10.1016/j.jct.2014.12.025>
<https://www.doi.org/10.1021/acs.jced.5b00838>
https://en.wikipedia.org/wiki/Joback_method
<http://link.springer.com/article/10.1007/BF02311772>
<http://webbook.nist.gov/cgi/cbook.cgi?ID=C18641822&Units=SI>
<http://pubs.acs.org/doi/abs/10.1021/ci9903071>

Legend

cpg:	Ideal gas heat capacity
dvisc:	Dynamic viscosity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
hvapt:	Enthalpy of vaporization at a given temperature
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rho:	Liquid Density
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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