

# DL-Glyceraldehyde, dimethyl ether

<b>Other names:</b>	2,3-Dimethoxypropanal
<b>Inchi:</b>	InChI=1S/C5H10O3/c1-7-4-5(3-6)8-2/h3,5H,4H2,1-2H3
<b>InchiKey:</b>	PNENILZGBBXDAG-UHFFFAOYSA-N
<b>Formula:</b>	C5H10O3
<b>SMILES:</b>	COCC(C=O)OC
<b>Mol. weight [g/mol]:</b>	118.13

## Physical Properties

Property code	Value	Unit	Source
gf	-320.74	kJ/mol	Joback Method
hf	-501.83	kJ/mol	Joback Method
hfus	9.85	kJ/mol	Joback Method
hvap	37.88	kJ/mol	Joback Method
log10ws	0.52		Crippen Method
logp	-0.153		Crippen Method
mcvol	94.620	ml/mol	McGowan Method
pc	3673.09	kPa	Joback Method
rinpol	983.00		NIST Webbook
rinpol	983.00		NIST Webbook
tb	406.86	K	Joback Method
tc	584.53	K	Joback Method
tf	217.57	K	Joback Method
vc	0.362	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	186.47	J/molxK	406.86	Joback Method
cpg	194.76	J/molxK	436.47	Joback Method
cpg	202.86	J/molxK	466.08	Joback Method
cpg	210.76	J/molxK	495.69	Joback Method
cpg	218.46	J/molxK	525.30	Joback Method
cpg	225.94	J/molxK	554.92	Joback Method
cpg	233.20	J/molxK	584.53	Joback Method

dvisc	0.0037241	Paxs	217.57	Joback Method
dvisc	0.0017929	Paxs	249.12	Joback Method
dvisc	0.0010173	Paxs	280.67	Joback Method
dvisc	0.0006473	Paxs	312.22	Joback Method
dvisc	0.0004475	Paxs	343.76	Joback Method
dvisc	0.0003292	Paxs	375.31	Joback Method
dvisc	0.0002539	Paxs	406.86	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U332933&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U332933&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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