

# 2,3,4,5,6-Pentabromobenzyl alcohol, n-propyl ether

Inchi:	InChI=1S/C10H9Br5O/c1-2-3-16-4-5-6(11)8(13)10(15)9(14)7(5)12/h2-4H2,1H3
InchiKey:	BUKHQPTWTSBKX-UHFFFAOYSA-N
Formula:	C10H9Br5O
SMILES:	CCCOc1c(Br)c(Br)c(Br)c(Br)c1Br
Mol. weight [g/mol]:	544.70

## Physical Properties

Property code	Value	Unit	Source
gf	64.18	kJ/mol	Joback Method
hf	-71.12	kJ/mol	Joback Method
hfus	41.37	kJ/mol	Joback Method
hvap	78.02	kJ/mol	Joback Method
log10ws	-8.51		Crippen Method
logp	6.426		Crippen Method
mcvol	221.370	ml/mol	McGowan Method
pc	3935.71	kPa	Joback Method
rinsol	2621.00		NIST Webbook
tb	833.00	K	Joback Method
tc	1099.01	K	Joback Method
tf	612.71	K	Joback Method
vc	0.816	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	436.54	J/molxK	833.00	Joback Method
cpg	444.87	J/molxK	877.34	Joback Method
cpg	452.67	J/molxK	921.67	Joback Method
cpg	460.01	J/molxK	966.01	Joback Method
cpg	466.95	J/molxK	1010.34	Joback Method
cpg	473.56	J/molxK	1054.68	Joback Method
cpg	479.89	J/molxK	1099.01	Joback Method
dvisc	0.0002979	Paxs	612.71	Joback Method
dvisc	0.0002349	Paxs	649.43	Joback Method

dvisc	0.0001900	Paxs	686.14	Joback Method
dvisc	0.0001570	Paxs	722.86	Joback Method
dvisc	0.0001322	Paxs	759.57	Joback Method
dvisc	0.0001130	Paxs	796.29	Joback Method
dvisc	0.0000980	Paxs	833.00	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U375315&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U375315&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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