

# Isophthalic acid, butyl pentachlorophenyl ester

Inchi:	InChI=1S/C18H13Cl5O4/c1-2-3-7-26-17(24)9-5-4-6-10(8-9)18(25)27-16-14(22)12(20)11
InchiKey:	YHXRPAPWRHPVTM-UHFFFAOYSA-N
Formula:	C18H13Cl5O4
SMILES:	CCCCOC(=O)c1cccc(C(=O)Oc2c(Cl)c(Cl)c(Cl)c(Cl)c2Cl)c1
Mol. weight [g/mol]:	470.56

## Physical Properties

Property code	Value	Unit	Source
gf	-259.77	kJ/mol	Joback Method
hf	-578.91	kJ/mol	Joback Method
hfus	54.68	kJ/mol	Joback Method
hvap	104.42	kJ/mol	Joback Method
log10ws	-8.48		Crippen Method
logp	7.130		Crippen Method
mcvol	293.040	ml/mol	McGowan Method
pc	1643.09	kPa	Joback Method
rinpol	3220.00		NIST Webbook
tb	1034.21	K	Joback Method
tc	1282.63	K	Joback Method
tf	714.50	K	Joback Method
vc	1.121	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	755.45	J/molxK	1034.21	Joback Method
cpg	774.80	J/molxK	1241.23	Joback Method
cpg	773.55	J/molxK	1199.82	Joback Method
cpg	771.00	J/molxK	1158.42	Joback Method
cpg	767.14	J/molxK	1117.02	Joback Method
cpg	761.96	J/molxK	1075.61	Joback Method
cpg	774.76	J/molxK	1282.63	Joback Method
dvisc	0.0000382	Paxs	1034.21	Joback Method
dvisc	0.0000456	Paxs	980.92	Joback Method

dvisc	0.0000555	Paxs	927.64	Joback Method
dvisc	0.0000692	Paxs	874.36	Joback Method
dvisc	0.0000888	Paxs	821.07	Joback Method
dvisc	0.0001180	Paxs	767.78	Joback Method
dvisc	0.0001635	Paxs	714.50	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U356576&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U356576&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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