

# Fumaric acid, butyl pent-4-enyl ester

<b>Inchi:</b>	InChI=1S/C13H20O4/c1-3-5-7-11-17-13(15)9-8-12(14)16-10-6-4-2/h3,8-9H,1,4-7,10-11H
<b>InchiKey:</b>	UPWPLXAPKLFNSX-CMDGGOBGSA-N
<b>Formula:</b>	C13H20O4
<b>SMILES:</b>	C=CCCCOC(=O)C=CC(=O)OCCCC
<b>Mol. weight [g/mol]:</b>	240.30

## Physical Properties

Property code	Value	Unit	Source
gf	-241.20	kJ/mol	Joback Method
hf	-558.60	kJ/mol	Joback Method
hfus	33.92	kJ/mol	Joback Method
hvap	62.13	kJ/mol	Joback Method
log10ws	-2.70		Crippen Method
logp	2.395		Crippen Method
mcvol	200.310	ml/mol	McGowan Method
pc	1923.67	kPa	Joback Method
rinsol	1679.00		NIST Webbook
tb	650.26	K	Joback Method
tc	835.14	K	Joback Method
tf	373.75	K	Joback Method
vc	0.772	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	524.33	J/molxK	650.26	Joback Method
cpg	538.41	J/molxK	681.07	Joback Method
cpg	551.80	J/molxK	711.89	Joback Method
cpg	564.51	J/molxK	742.70	Joback Method
cpg	576.55	J/molxK	773.51	Joback Method
cpg	587.94	J/molxK	804.33	Joback Method
cpg	598.69	J/molxK	835.14	Joback Method
dvisc	0.0014567	Paxs	373.75	Joback Method
dvisc	0.0007674	Paxs	419.83	Joback Method

dvisc	0.0004590	Paxs	465.92	Joback Method
dvisc	0.0003011	Paxs	512.00	Joback Method
dvisc	0.0002118	Paxs	558.09	Joback Method
dvisc	0.0001571	Paxs	604.17	Joback Method
dvisc	0.0001217	Paxs	650.26	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U348847&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U348847&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/16-165-1/Fumaric-acid-butyl-pent-4-enyl-ester.pdf>

Generated by Cheméo on 2024-04-26 17:22:15.718379709 +0000 UTC m=+16441384.638957021.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.