Tridecane, 2-methyl-

Other names:	2-Methyl-n-tridecane
	2-Methyltridecane
Inchi:	InChI=1S/C14H30/c1-4-5-6-7-8-9-10-11-12-13-14(2)3/h14H,4-13H2,1-3H3
InchiKey:	CJBFZKZYIPBBTO-UHFFFAOYSA-N
Formula:	C14H30
SMILES:	CCCCCCCCCCCC(C)C
Mol. weight [g/mol]:	198.39
CAS:	1560-96-9

Physical Properties

Property code	Value	Unit	Source
gf	64.56	kJ/mol	Joback Method
hf	-337.57	kJ/mol	Joback Method
hfus	28.49	kJ/mol	Joback Method
hvap	46.37	kJ/mol	Joback Method
log10ws	-5.44		Crippen Method
logp	5.563		Crippen Method
mcvol	208.120	ml/mol	McGowan Method
pc	1533.06	kPa	Joback Method
rinpol	1365.00		NIST Webbook
rinpol	1364.00		NIST Webbook
rinpol	1360.00		NIST Webbook
rinpol	1365.40		NIST Webbook
rinpol	1365.00		NIST Webbook
rinpol	1365.00		NIST Webbook
rinpol	1366.00		NIST Webbook
rinpol	1361.00		NIST Webbook
rinpol	1364.21		NIST Webbook
rinpol	1365.00		NIST Webbook
rinpol	1364.00		NIST Webbook
rinpol	1364.21		NIST Webbook
rinpol	1364.21		NIST Webbook
rinpol	1365.08		NIST Webbook
rinpol	1365.01		NIST Webbook
rinpol	1364.94		NIST Webbook
rinpol	1366.00		NIST Webbook
rinpol	1364.00		NIST Webbook

rinpol	1362.00		NIST Webbook
tb	519.28	K	Joback Method
tc	681.76	K	Joback Method
tf	247.30 ± 2.00	K	NIST Webbook
tf	251.00 ± 4.00	K	NIST Webbook
vc	0.814	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	606.55	J/mol×K	681.76	Joback Method
cpg	591.46	J/mol×K	654.68	Joback Method
cpg	575.74	J/mol×K	627.60	Joback Method
cpg	559.36	J/mol×K	600.52	Joback Method
cpg	542.30	J/mol×K	573.44	Joback Method
cpg	524.54	J/mol×K	546.36	Joback Method
cpg	506.07	J/mol×K	519.28	Joback Method
dvisc	0.0090478	Pa×s	232.54	Joback Method
dvisc	0.0001730	Pa×s	519.28	Joback Method
dvisc	0.0002394	Pa×s	471.49	Joback Method
dvisc	0.0003567	Pa×s	423.70	Joback Method
dvisc	0.0005882	Pa×s	375.91	Joback Method
dvisc	0.0011218	Pa×s	328.12	Joback Method
dvisc	0.0026666	Pa×s	280.33	Joback Method
hvapt	56.30	kJ/mol	459.00	NIST Webbook

Correlations

Information	Value
Property code	pvap
Equation	ln(Pvp) = A + B/(T + C)
Coeff. A	1.42606e+01
Coeff. B	-4.00753e+03
Coeff. C	-1.05430e+02
Temperature range (K), min.	392.23
Temperature range (K), max.	553.24

Sources

Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C1560969&Units=SI
The Yaws Handbook of Vapor Pressure:	https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure

Legend

cpg:	Ideal gas heat capacity
dvisc:	Dynamic viscosity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
hvapt:	Enthalpy of vaporization at a given temperature
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
pvap:	Vapor pressure
rinpol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume