

# 4-methylcara-3(10),4-diene

<b>Inchi:</b>	InChI=1S/C11H16/c1-7-5-9-10(6-8(7)2)11(9,3)4/h6,9-10H,1,5H2,2-4H3
<b>InchiKey:</b>	VCNGYUPLRXURU-UHFFFAOYSA-N
<b>Formula:</b>	C11H16
<b>SMILES:</b>	<chem>C=C1CC2C(C=C1C)C2(C)C</chem>
<b>Mol. weight [g/mol]:</b>	148.24

## Physical Properties

Property code	Value	Unit	Source
gf	211.35	kJ/mol	Joback Method
hf	-5.48	kJ/mol	Joback Method
hfus	12.86	kJ/mol	Joback Method
hvap	39.73	kJ/mol	Joback Method
log10ws	-3.20		Crippen Method
logp	3.165		Crippen Method
mcvol	135.530	ml/mol	McGowan Method
pc	2665.27	kPa	Joback Method
rinsol	1076.00		NIST Webbook
tb	467.70	K	Joback Method
tc	676.02	K	Joback Method
tf	292.71	K	Joback Method
vc	0.524	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	303.38	J/mol×K	467.70	Joback Method
cpg	320.63	J/mol×K	502.42	Joback Method
cpg	336.66	J/mol×K	537.14	Joback Method
cpg	351.59	J/mol×K	571.86	Joback Method
cpg	365.56	J/mol×K	606.58	Joback Method
cpg	378.68	J/mol×K	641.30	Joback Method
cpg	391.10	J/mol×K	676.02	Joback Method

# Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R492183&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R492183&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307I">http://pubs.acs.org/doi/abs/10.1021/ci990307I</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvac:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/16-253-3/4-methylcara-3-10-4-diene.pdf>

Generated by Cheméo on 2024-04-25 17:36:07.881058594 +0000 UTC m=+16355816.801635916.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.