

# (1R)-2,6,6-Trimethylbicyclo[3.1.1]hept-2-ene

<b>Other names:</b>	1R-«alpha»-Pinene Bicyclo[3.1.1]hept-2-ene, 2,6,6-trimethyl-, (1R)- 2,6,6-Trimethylbicyclo[3.1.1]hept-2-ene-, (1R,5R)- d-«alpha»-Pinene 1R-(+)-«alpha»-Pinene (R)-«alpha»-Pinene «alpha»-Pinene, (D)- Dextro-alpha-pinene (+)-pin-2(3)-ene «alpha»-Pinene(dextro)
<b>Inchi:</b>	InChI=1S/C10H16/c1-7-4-5-8-6-9(7)10(8,2)3/h4,8-9H,5-6H2,1-3H3/t8-,9?/m1/s1
<b>InchiKey:</b>	GRWFGVWFFZKLTl-VEDVMXKPSA-N
<b>Formula:</b>	C10H16
<b>SMILES:</b>	CC1=CCC2CC1C2(C)C
<b>Mol. weight [g/mol]:</b>	136.23
<b>CAS:</b>	7785-70-8

## Physical Properties

Property code	Value	Unit	Source
chl	-6156.80	kJ/mol	NIST Webbook
gf	149.85	kJ/mol	Joback Method
hf	-69.08	kJ/mol	Joback Method
hfus	11.43	kJ/mol	Joback Method
hvap	37.35	kJ/mol	Joback Method
log10ws	-2.93		Crippen Method
logp	2.999		Crippen Method
mcvol	125.740	ml/mol	McGowan Method
pc	2890.51	kPa	Joback Method
rinpol	937.00		NIST Webbook
rinpol	929.00		NIST Webbook
rinpol	929.00		NIST Webbook
rinpol	922.00		NIST Webbook
rinpol	937.00		NIST Webbook
rinpol	922.00		NIST Webbook
rinpol	935.00		NIST Webbook
ripol	1010.00		NIST Webbook
ripol	1016.00		NIST Webbook

tb	428.20	K	NIST Webbook
tc	654.83	K	Joback Method
tf	267.76	K	Joback Method
vc	0.484	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	274.36	J/mol×K	445.66	Joback Method
cpg	292.47	J/mol×K	480.52	Joback Method
cpg	309.23	J/mol×K	515.38	Joback Method
cpg	324.78	J/mol×K	550.25	Joback Method
cpg	339.25	J/mol×K	585.11	Joback Method
cpg	352.77	J/mol×K	619.97	Joback Method
cpg	365.49	J/mol×K	654.83	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C7785708&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C7785708&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>chl:</b>	Standard liquid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume

<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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