

# Benzene, 1,3,5-trimethyl-2-(1-chloro-1-methylethyl)

Inchi:	InChI=1S/C12H17Cl/c1-8-6-9(2)11(10(3)7-8)12(4,5)13/h6-7H,1-5H3
InchiKey:	SEPLMBZJYZUXJF-UHFFFAOYSA-N
Formula:	C12H17Cl
SMILES:	Cc1cc(C)c(C(C)(C)Cl)c(C)c1
Mol. weight [g/mol]:	196.72

## Physical Properties

Property code	Value	Unit	Source
gf	124.59	kJ/mol	Joback Method
hf	-113.38	kJ/mol	Joback Method
hfus	16.49	kJ/mol	Joback Method
hvap	49.66	kJ/mol	Joback Method
log10ws	-4.44		Crippen Method
logp	4.086		Crippen Method
mcvol	168.420	ml/mol	McGowan Method
pc	2263.26	kPa	Joback Method
rinsol	1389.00		NIST Webbook
tb	549.78	K	Joback Method
tc	770.39	K	Joback Method
tf	321.32	K	Joback Method
vc	0.637	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	379.95	J/molxK	549.78	Joback Method
cpg	396.03	J/molxK	586.55	Joback Method
cpg	411.12	J/molxK	623.32	Joback Method
cpg	425.28	J/molxK	660.09	Joback Method
cpg	438.55	J/molxK	696.85	Joback Method
cpg	450.98	J/molxK	733.62	Joback Method
cpg	462.63	J/molxK	770.39	Joback Method
dvisc	0.0016810	Paxs	321.32	Joback Method
dvisc	0.0009497	Paxs	359.40	Joback Method

dvisc	0.0005985	Paxs	397.47	Joback Method
dvisc	0.0004089	Paxs	435.55	Joback Method
dvisc	0.0002970	Paxs	473.63	Joback Method
dvisc	0.0002263	Paxs	511.70	Joback Method
dvisc	0.0001790	Paxs	549.78	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R131665&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R131665&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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<https://www.chemeo.com/cid/16-319-0/Benzene-1-3-5-trimethyl-2-1-chloro-1-methylethyl.pdf>

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