

# 2,4-Dinitrophenylmethylnitramine

<b>Inchi:</b>	InChI=1S/C7H6N4O6/c1-8(11(16)17)6-3-2-5(9(12)13)4-7(6)10(14)15/h2-4H,1H3
<b>InchiKey:</b>	BUGRLMJGTJHURD-UHFFFAOYSA-N
<b>Formula:</b>	C7H6N4O6
<b>SMILES:</b>	CN(c1ccc([N+](=O)[O-])cc1[N+](=O)[O-])[N+](=O)[O-]
<b>Mol. weight [g/mol]:</b>	242.15
<b>CAS:</b>	19092-03-6

## Physical Properties

Property code	Value	Unit	Source
chs	-3628.20 ± 3.70	kJ/mol	NIST Webbook
gf	318.64	kJ/mol	Joback Method
hf	61.03	kJ/mol	Joback Method
hfs	16.20 ± 3.70	kJ/mol	NIST Webbook
hfus	44.25	kJ/mol	Joback Method
hvap	86.59	kJ/mol	Joback Method
log10ws	-3.34		Crippen Method
logp	1.131		Crippen Method
mcvol	147.970	ml/mol	McGowan Method
pc	4135.61	kPa	Joback Method
tb	864.16	K	Joback Method
tc	1147.85	K	Joback Method
tf	683.41	K	Joback Method
vc	0.584	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	412.43	J/mol×K	864.16	Joback Method
cpg	420.28	J/mol×K	911.44	Joback Method
cpg	427.21	J/mol×K	958.72	Joback Method
cpg	433.31	J/mol×K	1006.01	Joback Method
cpg	438.67	J/mol×K	1053.29	Joback Method
cpg	443.38	J/mol×K	1100.57	Joback Method
cpg	447.54	J/mol×K	1147.85	Joback Method

# Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C19092036&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C19092036&amp;Units=SI</a>

# Legend

<b>chs:</b>	Standard solid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfs:</b>	Solid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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