

# cis-2,5-Diisopropyl-1,3-dioxane

|                      |                                                                            |
|----------------------|----------------------------------------------------------------------------|
| Inchi:               | InChI=1S/C10H20O2/c1-7(2)9-5-11-10(8(3)4)12-6-9/h7-10H,5-6H2,1-4H3/t9-,10+ |
| InchiKey:            | Y AISJNQHUBKIPN-AOOOYVTPSA-N                                               |
| Formula:             | C10H20O2                                                                   |
| SMILES:              | CC(C)C1COC(C(C)C)OC1                                                       |
| Mol. weight [g/mol]: | 172.26                                                                     |
| CAS:                 | 25925-00-2                                                                 |

## Physical Properties

| Property code | Value            | Unit    | Source         |
|---------------|------------------|---------|----------------|
| chl           | -6169.00 ± 18.00 | kJ/mol  | NIST Webbook   |
| gf            | -127.06          | kJ/mol  | Joback Method  |
| hf            | -490.31          | kJ/mol  | Joback Method  |
| hfl           | -625.00 ± 18.00  | kJ/mol  | NIST Webbook   |
| hfus          | 23.47            | kJ/mol  | Joback Method  |
| hvap          | 46.22            | kJ/mol  | Joback Method  |
| log10ws       | -1.96            |         | Crippen Method |
| logp          | 2.288            |         | Crippen Method |
| mcvol         | 152.640          | ml/mol  | McGowan Method |
| pc            | 2480.12          | kPa     | Joback Method  |
| tb            | 496.10           | K       | Joback Method  |
| tc            | 700.34           | K       | Joback Method  |
| tf            | 228.74           | K       | Joback Method  |
| vc            | 0.557            | m3/kmol | Joback Method  |

## Temperature Dependent Properties

| Property code | Value  | Unit    | Temperature [K] | Source        |
|---------------|--------|---------|-----------------|---------------|
| cpg           | 368.52 | J/mol×K | 496.10          | Joback Method |
| cpg           | 388.02 | J/mol×K | 530.14          | Joback Method |
| cpg           | 406.58 | J/mol×K | 564.18          | Joback Method |
| cpg           | 424.19 | J/mol×K | 598.22          | Joback Method |
| cpg           | 440.89 | J/mol×K | 632.26          | Joback Method |
| cpg           | 456.68 | J/mol×K | 666.30          | Joback Method |
| cpg           | 471.58 | J/mol×K | 700.34          | Joback Method |

|       |           |      |        |               |
|-------|-----------|------|--------|---------------|
| dvisc | 0.0134801 | Pa×s | 228.74 | Joback Method |
| dvisc | 0.0041346 | Pa×s | 273.30 | Joback Method |
| dvisc | 0.0017664 | Pa×s | 317.86 | Joback Method |
| dvisc | 0.0009302 | Pa×s | 362.42 | Joback Method |
| dvisc | 0.0005637 | Pa×s | 406.98 | Joback Method |
| dvisc | 0.0003771 | Pa×s | 451.54 | Joback Method |
| dvisc | 0.0002711 | Pa×s | 496.10 | Joback Method |

## Sources

|                        |                                                                                                                                               |
|------------------------|-----------------------------------------------------------------------------------------------------------------------------------------------|
| <b>McGowan Method:</b> | <a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>                         |
| <b>NIST Webbook:</b>   | <a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C25925002&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C25925002&amp;Units=SI</a> |
| <b>Crippen Method:</b> | <a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>                                     |
| <b>Crippen Method:</b> | <a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>                             |
| <b>Joback Method:</b>  | <a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>                                         |

## Legend

|                 |                                                           |
|-----------------|-----------------------------------------------------------|
| <b>chl:</b>     | Standard liquid enthalpy of combustion                    |
| <b>cpg:</b>     | Ideal gas heat capacity                                   |
| <b>dvisc:</b>   | Dynamic viscosity                                         |
| <b>gf:</b>      | Standard Gibbs free energy of formation                   |
| <b>hf:</b>      | Enthalpy of formation at standard conditions              |
| <b>hfl:</b>     | Liquid phase enthalpy of formation at standard conditions |
| <b>hfus:</b>    | Enthalpy of fusion at standard conditions                 |
| <b>hvap:</b>    | Enthalpy of vaporization at standard conditions           |
| <b>log10ws:</b> | Log10 of Water solubility in mol/l                        |
| <b>logp:</b>    | Octanol/Water partition coefficient                       |
| <b>mcvol:</b>   | McGowan's characteristic volume                           |
| <b>pc:</b>      | Critical Pressure                                         |
| <b>tb:</b>      | Normal Boiling Point Temperature                          |
| <b>tc:</b>      | Critical Temperature                                      |
| <b>tf:</b>      | Normal melting (fusion) point                             |
| <b>vc:</b>      | Critical Volume                                           |

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