

4-Octanol, 4-methyl-

Other names:	4-Methyl-4-octanol
Inchi:	InChI=1S/C9H20O/c1-4-6-8-9(3,10)7-5-2/h10H,4-8H2,1-3H3
InchiKey:	RXSIKQJQLQRQQY-UHFFFAOYSA-N
Formula:	C9H20O
SMILES:	CCCCC(C)(O)CCC
Mol. weight [g/mol]:	144.25
CAS:	23418-37-3

Physical Properties

Property code	Value	Unit	Source
gf	-109.08	kJ/mol	Joback Method
hf	-390.07	kJ/mol	Joback Method
hfus	15.74	kJ/mol	Joback Method
hvap	51.01	kJ/mol	Joback Method
log10ws	-2.97		Crippen Method
logp	2.728		Crippen Method
mcvol	143.540	ml/mol	McGowan Method
pc	2574.11	kPa	Joback Method
tb	453.15 ± 3.00	K	NIST Webbook
tb	453.65 ± 3.00	K	NIST Webbook
tc	662.00	K	Joback Method
tf	254.43	K	Joback Method
vc	0.547	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	337.27	J/mol×K	494.27	Joback Method
cpg	350.61	J/mol×K	522.22	Joback Method
cpg	363.35	J/mol×K	550.18	Joback Method
cpg	375.49	J/mol×K	578.13	Joback Method
cpg	387.08	J/mol×K	606.09	Joback Method
cpg	398.12	J/mol×K	634.04	Joback Method
cpg	408.65	J/mol×K	662.00	Joback Method

dvisc	0.0562745	Pa×s	254.43	Joback Method
dvisc	0.0106579	Pa×s	294.40	Joback Method
dvisc	0.0030048	Pa×s	334.38	Joback Method
dvisc	0.0011101	Pa×s	374.35	Joback Method
dvisc	0.0004970	Pa×s	414.32	Joback Method
dvisc	0.0002563	Pa×s	454.30	Joback Method
dvisc	0.0001471	Pa×s	494.27	Joback Method

Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.65335e+01
Coeff. B	-4.59623e+03
Coeff. C	-6.74040e+01
Temperature range (K), min.	350.32
Temperature range (K), max.	476.98

Sources

Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C23418373&Units=SI
The Yaws Handbook of Vapor Pressure:	https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l

Legend

cpg:	Ideal gas heat capacity
dvisc:	Dynamic viscosity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions

h_{vap}:	Enthalpy of vaporization at standard conditions
log₁₀w_s:	Log10 of Water solubility in mol/l
log_p:	Octanol/Water partition coefficient
mc_{vol}:	McGowan's characteristic volume
p_c:	Critical Pressure
p_{vap}:	Vapor pressure
t_b:	Normal Boiling Point Temperature
t_c:	Critical Temperature
t_f:	Normal melting (fusion) point
v_c:	Critical Volume

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