

Ethane, 1,1-dibromo-

Other names:	1,1-Dibromoethane CH ₃ CHBr ₂ Dibromoethane ETHYLIDENE DIBROMIDE Ethylidene bromide
Inchi:	InChI=1S/C2H4Br2/c1-2(3)4/h2H,1H3
InchiKey:	APQIUTYORBAGEZ-UHFFFAOYSA-N
Formula:	C ₂ H ₄ Br ₂
SMILES:	CC(Br)Br
Mol. weight [g/mol]:	187.86
CAS:	557-91-5

Physical Properties

Property code	Value	Unit	Source
gf	-7.84	kJ/mol	Joback Method
hf	-41.00	kJ/mol	NIST Webbook
hfus	7.98	kJ/mol	Joback Method
hvap	32.53	kJ/mol	Joback Method
ie	10.19 ± 0.03	eV	NIST Webbook
ie	10.17	eV	NIST Webbook
ie	10.17	eV	NIST Webbook
log10ws	-2.13		Crippen Method
logp	2.122		Crippen Method
mcvol	74.040	ml/mol	McGowan Method
pc	6259.01	kPa	Joback Method
rinpol	712.00		NIST Webbook
rinpol	718.00		NIST Webbook
rinpol	698.00		NIST Webbook
rinpol	706.00		NIST Webbook
rinpol	724.00		NIST Webbook
tb	382.65 ± 0.70	K	NIST Webbook
tb	381.20	K	NIST Webbook
tc	592.98	K	Joback Method
tf	216.90	K	Joback Method
vc	0.266	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	91.34	J/molxK	377.04	Joback Method
cpg	111.78	J/molxK	556.99	Joback Method
cpg	108.28	J/molxK	521.00	Joback Method
cpg	104.51	J/molxK	485.01	Joback Method
cpg	100.45	J/molxK	449.02	Joback Method
cpg	96.07	J/molxK	413.03	Joback Method
cpg	115.02	J/molxK	592.98	Joback Method
dvisc	0.0004806	Paxs	377.04	Joback Method
dvisc	0.0006009	Paxs	350.35	Joback Method
dvisc	0.0007796	Paxs	323.66	Joback Method
dvisc	0.0010599	Paxs	296.97	Joback Method
dvisc	0.0015312	Paxs	270.28	Joback Method
dvisc	0.0023976	Paxs	243.59	Joback Method
dvisc	0.0041922	Paxs	216.90	Joback Method

Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.40647e+01
Coeff. B	-3.06198e+03
Coeff. C	-5.70550e+01
Temperature range (K), min.	279.31
Temperature range (K), max.	406.87

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/T + C \cdot \ln(T) + D \cdot T^2$
Coeff. A	5.57186e+01
Coeff. B	-6.50116e+03
Coeff. C	-5.75296e+00
Coeff. D	1.01867e-06
Temperature range (K), min.	210.15

Sources

Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
KDB:	https://www.chemeo.com/files/research/kdb/mol/mol1570.mol
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C557915&Units=SI
The Yaws Handbook of Vapor Pressure:	https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure
KDB Vapor Pressure Data:	https://www.chemeo.com/research/kdb/hcprop/showprop.php?cmpid=1570
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071

Legend

cpg:	Ideal gas heat capacity
dvisc:	Dynamic viscosity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
ie:	Ionization energy
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
pvap:	Vapor pressure
rinpolar:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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