

# p-Benzoquinone, 2-methyl-

<b>Other names:</b>	2,5-Cyclohexadiene-1,4-dione, 2-methyl-p-Toluquinone Methyl-p-benzoquinone Methyl-1,4-benzoquinone Toluquinone Tolyquinone 2-Methyl-p-benzoquinone 2-Methyl-1,4-benzoquinone 2-Methylbenzoquinone 2-Methylquinone 1,4-Toluquinone 2-Methyl-1,4-quinone 2-Methylbenzoquinone-1,4 2-Methyl-1,4-benzochinon 1,4-Toluchinon 2-Methyl-2,5-cyclohexadiene-1,4-dione NSC 405002
<b>Inchi:</b>	InChI=1S/C7H6O2/c1-5-4-6(8)2-3-7(5)9/h2-4H,1H3
<b>InchiKey:</b>	VTWDKFNVVLAELH-UHFFFAOYSA-N
<b>Formula:</b>	C7H6O2
<b>SMILES:</b>	CC1=CC(=O)C=CC1=O
<b>Mol. weight [g/mol]:</b>	122.12
<b>CAS:</b>	553-97-9

## Physical Properties

Property code	Value	Unit	Source
chs	-3361.00	kJ/mol	NIST Webbook
ea	1.85 ± 0.06	eV	NIST Webbook
ea	1.85 ± 0.05	eV	NIST Webbook
gf	-154.67	kJ/mol	Joback Method
hf	-284.46	kJ/mol	Joback Method
hfus	5.73	kJ/mol	Joback Method
hvap	41.65	kJ/mol	Joback Method
ie	9.78 ± 0.02	eV	NIST Webbook
ie	9.78	eV	NIST Webbook
log10ws	-0.92		Crippen Method
logp	0.641		Crippen Method

mvol	93.170	ml/mol	McGowan Method
pc	4299.92	kPa	Joback Method
rinpol	1008.00		NIST Webbook
rinpol	1031.00		NIST Webbook
rinpol	1015.00		NIST Webbook
rinpol	1018.00		NIST Webbook
rinpol	1025.00		NIST Webbook
ripol	1624.00		NIST Webbook
ripol	1616.00		NIST Webbook
ripol	1640.00		NIST Webbook
ripol	1633.00		NIST Webbook
tb	522.72	K	Joback Method
tc	771.09	K	Joback Method
tf	341.15 ± 2.00	K	NIST Webbook
vc	0.347	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	194.24	J/mol×K	522.72	Joback Method
cpg	206.10	J/mol×K	564.11	Joback Method
cpg	217.49	J/mol×K	605.51	Joback Method
cpg	228.34	J/mol×K	646.90	Joback Method
cpg	238.62	J/mol×K	688.30	Joback Method
cpg	248.26	J/mol×K	729.69	Joback Method
cpg	257.20	J/mol×K	771.09	Joback Method

## Sources

**Crippen Method:**

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

**Joback Method:**

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

**McGowan Method:**

<http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:**

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C553979&Units=SI>

**Crippen Method:**

<http://pubs.acs.org/doi/abs/10.1021/ci9903071>

# Legend

<b>chs:</b>	Standard solid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>ea:</b>	Electron affinity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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