

# Benzoic acid, 4-nitro-, phenylmethyl ester

<b>Other names:</b>	Benzoic acid, p-nitro-, benzyl ester Benzyl p-nitrobenzoate Benzyl 4-nitrobenzoate 4-Nitrobenzoic acid, benzyl ester Benzoic acid, 4-nitro-, benzyl ester
<b>Inchi:</b>	InChI=1S/C14H11NO4/c16-14(19-10-11-4-2-1-3-5-11)12-6-8-13(9-7-12)15(17)18/h1-9H,
<b>InchiKey:</b>	XCQFGJYVKKCUFX-UHFFFAOYSA-N
<b>Formula:</b>	C14H11NO4
<b>SMILES:</b>	O=C(OCc1ccccc1)c1ccc([N+](=O)[O-])cc1
<b>Mol. weight [g/mol]:</b>	257.24
<b>CAS:</b>	14786-27-7

## Physical Properties

Property code	Value	Unit	Source
gf	83.82	kJ/mol	Joback Method
hf	-126.26	kJ/mol	Joback Method
hfus	33.86	kJ/mol	Joback Method
hvap	77.72	kJ/mol	Joback Method
log10ws	-4.47		Crippen Method
logp	2.952		Crippen Method
mcvol	185.460	ml/mol	McGowan Method
pc	2899.85	kPa	Joback Method
rinpol	2087.00		NIST Webbook
rinpol	2071.00		NIST Webbook
rinpol	2071.00		NIST Webbook
rinpol	2087.00		NIST Webbook
rinpol	2106.00		NIST Webbook
rinpol	2121.00		NIST Webbook
rinpol	2141.00		NIST Webbook
rinpol	2087.00		NIST Webbook
rinpol	2097.00		NIST Webbook
rinpol	2106.00		NIST Webbook
rinpol	2097.00		NIST Webbook
tb	806.19	K	Joback Method
tc	1067.65	K	Joback Method
tf	528.67	K	Joback Method
vc	0.710	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	510.55	J/mol×K	806.19	Joback Method
cpg	522.67	J/mol×K	849.77	Joback Method
cpg	533.54	J/mol×K	893.34	Joback Method
cpg	543.22	J/mol×K	936.92	Joback Method
cpg	551.78	J/mol×K	980.50	Joback Method
cpg	559.28	J/mol×K	1024.08	Joback Method
cpg	565.80	J/mol×K	1067.65	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C14786277&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C14786277&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpolar:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point

**vc:** Critical Volume

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