

1,2-DIMETHOXY-3,4-DICHLORO-BENZENE

Other names:	Benzene, 3,4-dichloro-1,2-dimethoxy
Inchi:	InChI=1S/C8H8Cl2O2/c1-11-6-4-3-5(9)7(10)8(6)12-2/h3-4H,1-2H3
InchiKey:	WEMHPBMPYVGSJI-UHFFFAOYSA-N
Formula:	C8H8Cl2O2
SMILES:	COc1ccc(Cl)c(Cl)c1OC
Mol. weight [g/mol]:	207.06

Physical Properties

Property code	Value	Unit	Source
hf	-268.39	kJ/mol	Cheméo Relay (1.0)
hfus	20.12	kJ/mol	Joback Method
hvap	70.73	kJ/mol	Cheméo Relay (1.0)
ie	8.19	eV	Cheméo Relay (1.0)
log10ws	-3.22		Cheméo Relay (1.0)
logp	3.011		Crippen Method
mcvol	136.040	ml/mol	McGowan Method
rinpol	1455.00		NIST Webbook
rinpol	1448.00		NIST Webbook
rinpol	1451.00		NIST Webbook
rinpol	1477.00		NIST Webbook
rinpol	1461.00		NIST Webbook
ripol	2108.00		NIST Webbook
ripol	2127.00		NIST Webbook
ripol	2150.00		NIST Webbook
ripol	2171.00		NIST Webbook
ripol	2134.00		NIST Webbook
tb	533.55	K	Cheméo Relay (1.0)
tf	328.70	K	Cheméo Relay (1.0)

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	268.48	J/mol×K	543.76	Joback Method
cpg	321.89	J/mol×K	765.36	Joback Method

cpg	314.19	J/mol×K	728.42	Joback Method
cpg	305.98	J/mol×K	691.49	Joback Method
cpg	297.29	J/mol×K	654.56	Joback Method
cpg	288.14	J/mol×K	617.63	Joback Method
cpg	278.53	J/mol×K	580.69	Joback Method
dvisc	0.0003249	Pa×s	445.98	Joback Method
dvisc	0.0001753	Pa×s	543.76	Joback Method
dvisc	0.0002098	Pa×s	511.17	Joback Method
dvisc	0.0002572	Pa×s	478.57	Joback Method
dvisc	0.0008515	Pa×s	348.20	Joback Method
dvisc	0.0004258	Pa×s	413.39	Joback Method
dvisc	0.0005846	Pa×s	380.79	Joback Method

Sources

Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Cheméo Relay (1.0):	
Cheméo Relay (1.0, beta):	
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C90283004&Units=SI
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772

Legend

cpg:	Ideal gas heat capacity
dvisc:	Dynamic viscosity
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
ie:	Ionization energy
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
rinpol:	Non-polar retention indices
ripol:	Polar retention indices
tb:	Normal Boiling Point Temperature
tf:	Normal melting (fusion) point

Latest version available from:

<https://www.chemeo.com/cid/16-567-5/1-2-DIMETHOXY-3-4-DICHLORO-BENZENE.pdf>

Generated by Cheméo on 2026-07-20 15:35:43.228733081 +0000 UTC m=+72839.070618480.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.