

# 1,2,4-Trimethoxybenzene

<b>Other names:</b>	Benzene, 1,2,4-trimethoxy-
<b>Inchi:</b>	InChI=1S/C9H12O3/c1-10-7-4-5-8(11-2)9(6-7)12-3/h4-6H,1-3H3
<b>InchiKey:</b>	AGIQIOSHSMJYJP-UHFFFAOYSA-N
<b>Formula:</b>	C9H12O3
<b>SMILES:</b>	COc1ccc(OC)c(OC)c1
<b>Mol. weight [g/mol]:</b>	168.19
<b>CAS:</b>	135-77-3

## Physical Properties

Property code	Value	Unit	Source
gf	-196.95	kJ/mol	Joback Method
hf	-412.16	kJ/mol	Joback Method
hfus	15.89	kJ/mol	Joback Method
hvap	46.46	kJ/mol	Joback Method
ie	7.50 ± 0.15	eV	NIST Webbook
ie	7.49	eV	NIST Webbook
log10ws	-1.83		Crippen Method
logp	1.712		Crippen Method
mcvol	131.520	ml/mol	McGowan Method
pc	2979.54	kPa	Joback Method
rinpol	1331.30		NIST Webbook
rinpol	1374.00		NIST Webbook
rinpol	1377.80		NIST Webbook
ripol	2095.00		NIST Webbook
ripol	2133.00		NIST Webbook
ripol	2094.00		NIST Webbook
tb	520.20	K	NIST Webbook
tc	714.26	K	Joback Method
tf	309.34	K	Joback Method
vc	0.485	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
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cpg	285.91	J/molxK	509.22	Joback Method
cpg	298.31	J/molxK	543.39	Joback Method
cpg	310.28	J/molxK	577.57	Joback Method
cpg	321.81	J/molxK	611.74	Joback Method
cpg	332.87	J/molxK	645.92	Joback Method
cpg	343.44	J/molxK	680.09	Joback Method
cpg	353.50	J/molxK	714.26	Joback Method
dvisc	0.0008604	Paxs	309.34	Joback Method
dvisc	0.0005446	Paxs	342.65	Joback Method
dvisc	0.0003738	Paxs	375.97	Joback Method
dvisc	0.0002728	Paxs	409.28	Joback Method
dvisc	0.0002088	Paxs	442.59	Joback Method
dvisc	0.0001658	Paxs	475.91	Joback Method
dvisc	0.0001358	Paxs	509.22	Joback Method

## Sources

**McGowan Method:**

<http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:**

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C135773&Units=SI>

**Crippen Method:**

<http://pubs.acs.org/doi/abs/10.1021/ci9903071>

**Crippen Method:**

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

**Joback Method:**

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices

**tb:** Normal Boiling Point Temperature  
**tc:** Critical Temperature  
**tf:** Normal melting (fusion) point  
**vc:** Critical Volume

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