

# N,N'-[Methylenebis(2-chloro-4,1-phenylene)]bis(2,

**Other names:** Acetamide, N,N'-[methylenebis(2-chloro-4,1-phenylene)]bis[2,2,2-trifluoro-N,N'-[Methanediylbis(2-chlorobenzene-4,1-diyl)]bis(2,2,2-trifluoroacetamide)  
N-[2-Chloro-4-[[3-chloro-4-[(2,2,2-trifluoroacetyl)amino]phenyl]methyl]phenyl]-2,2,2-trifluoroacetamide

**Inchi:** InChI=1S/C17H10Cl2F6N2O2/c18-10-6-8(1-3-12(10)26-14(28)16(20,21)22)5-9-2-4-13(14,15,17)

**InchiKey:** XOAPUMLLYZSBIT-UHFFFAOYSA-N

**Formula:** C17H10Cl2F6N2O2

**SMILES:** O=C(Nc1ccc(Cc2ccc(NC(=O)C(F)(F)F)c(Cl)c2)cc1Cl)C(F)(F)F

**Mol. weight [g/mol]:** 459.17

## Physical Properties

Property code	Value	Unit	Source
gf	-987.54	kJ/mol	Joback Method
hf	-1310.89	kJ/mol	Joback Method
hfus	51.75	kJ/mol	Joback Method
hvap	88.28	kJ/mol	Joback Method
log10ws	-6.67		Crippen Method
logp	5.586		Crippen Method
mvol	261.070	ml/mol	McGowan Method
pc	1733.22	kPa	Joback Method
rinpol	2453.00		NIST Webbook
tb	933.74	K	Joback Method
tc	1155.29	K	Joback Method
tf	657.67	K	Joback Method
vc	1.038	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	746.44	J/molxK	933.74	Joback Method
cpg	754.76	J/molxK	970.66	Joback Method
cpg	762.40	J/molxK	1007.59	Joback Method
cpg	769.48	J/molxK	1044.51	Joback Method
cpg	776.10	J/molxK	1081.44	Joback Method
cpg	782.36	J/molxK	1118.36	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U373067&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U373067&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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